

Curriculum of Diploma Programme in Library & Information Science



**Department of Science, Technology and Technical
Education (DSTTE), Govt. of Bihar**

**State Board of Technical Education
(SBTE), Bihar**

Semester – I Teaching & Learning Scheme

Course Codes	Category of course	Course Titles	Teaching & Learning Scheme (Hours/Week)					
			Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
			L	T				
2400101	ASC	Basic Engg. Mathematics (ME, ME (Auto), CE, MIE, CSE, AIML, EE, CRE, CHE, ELX, ELX (R))	02	01	-	02	05	04
2400103A	ASC	Applied Chemistry–A (CE, ME, ME (Auto), MIE, AE, FTS, CRE, CHE)	03	-	04	02	09	06
2441103	BCC	Fundamentals of Library and information Science	03	-	04	02	09	06
2400104	HSC	Communication Skills (English) (Common for all Programmes)	03	-	04	02	09	06
2418104	BCC	Fundamental of IT System (CSE, AIML)	03	-	04	02	09	06
2400006	NRC	Environmental Education and Sustainable Development (Common for All Programmes)	01	-	01	01	03	02
Total			15	1	17	11	44	30

Note: Prefix will be added to course code if applicable (T for Theory Paper, P for Practical Paper and S for Term Work)

Legend:

- CI: Classroom Instruction (Includes different instructional/implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem based learning etc. to deliver theoretical concepts)
- LI: Laboratory Instruction (Includes experiments/practical performances /problem-based experiences in laboratory, work shop, field or other locations using different instructional/Implementation strategies)
- Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.
- TW: Term work (includes assignments, seminars, micro projects, industrial visits, any other student activities etc.)
- SL: Self Learning, MOOCs, spoken tutorials, online educational resources etc.
- C: Credits= (1x CI hours) + (0.5 x LI hours) + (0.5 x Notional hours)
- Note:** TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of teacher to ensure outcome of learning.

Semester - I Assessment Scheme

Course Codes	Category of course	Course Titles	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
			Theory Assessment (TA)		Term work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
			Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2400101	ASC	Basic Engg. Mathematics (ME, ME (Auto), CE, MIE, CSE, AIML, EE, CRE, CHE, ELX, ELX (R))	30	70	20	30	-	-	150
2400103A	ASC	Applied Chemistry-A (CE, ME, ME (Auto), MIE, AE, FTS, CRE, CHE)	30	70	20	30	20	30	200
2441103	BCC	Fundamentals of Library and information Science	30	70	20	30	20	30	200
2400104	HSC	Communication Skills (English) (Common for all Programmes)	30	70	20	30	20	30	200
2418104	BCC	Fundamental of IT System (CSE, AIML)	30	70	20	30	20	30	200
2400006	ASC	Environmental Education and Sustainable Development (Common for All Programmes)	15	-	10	-	10	15	50
Total			165	350	110	150	90	135	1000

Note: Prefix will be added to course code if applicable (T for Theory Paper, P for Practical Paper and S for Term Work)

Legend:

PTA: Progressive Theory Assessment in class room (includes class test, mid-term test and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro projects, industrial visits, self-learning, any other student activities etc.

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignment, micro project, seminar and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/ presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria of internal as well as external assessment may vary as per the requirement of respective course. For valid and reliable assessment, the internal faculty should prepare checklist & rubrics for these activities.

- A) **Course Code** : 2400101(T2400101/S2400101)
- B) **Course Title** : Basic Engg. Mathematics
(CE, ME, ME (Auto), CSE, EE, ELX, ELX (R), AIML, MIE, CRE, CHE)
- C) **Pre-requisite Course(s)** :
- D) **Rationale** :

This course is an extension of the course based on Mathematics of the first semester namely Basic Engineering Mathematics. The course is designed to inculcate its application in relevant branches of engineering and technology. With calculus, we can find how the changing conditions of a system affect us, and we can control a system. Definite integral is a powerful tool that helps us realize and model the world around us. Differential equations are widely applied to modern natural phenomena, engineering systems, and many other situations. Numerical methods offer approximate but credible accurate solutions to problems that are not readily or possibly solved by closed-form solution methods. On the other hand, Numerical integration is a computational (approximate) approach to evaluating definite integrals. It has a lot of applications in engineering such as in the computation of areas, volumes, and surfaces. It also has the advantage of being easily programmable in computer software. Probability distributions are useful for modeling, simulation, analysis, and inference on varieties of natural processes and physical phenomena. A situation in which an experiment is repeated a fixed number of times can be modeled, engineers need to apply existing knowledge of success and failure to a specific analytical scenario.

- E) **Course Outcomes (COs):** After the completion of the course, teachers are expected to ensure the accomplishment of the following course outcomes by the learners. For this, the learners are expected to perform various activities related to three learning domains (Cognitive, Psychomotor, and Affective) in classroom/ laboratory/ workshop/ field/ industry.

After completion of the course, the students will be able to-

- CO-1** Demonstrate the ability to solve engineering-related problems based on applications of algebra.
- CO-2** Use the concept of derivative as a tool to solve engineering-related problems.
- CO-3** Apply differential calculus to solve branch-specific problems.
- CO-4** Use the concept of Coordinate geometry to solve branch-specific engineering-related problems.
- CO-5** Apply techniques and methods of probability and statistics to crack branch-specific problems.

- F) **Suggested Course Articulation Matrix (CAM):**

Course Outcomes (COs)	Program Outcomes (POs)							Programme Specific Outcomes* (PSOs)	
	PO-1 Basic and Discipline Specific Knowledge	PO-2 Problem Analysis	PO-3 Design/ Development of Solutions	PO-4 Engineering Tools	PO-5 Engineering Practices for Society, Sustainability and Environment	PO-6 Project Management	PO-7 Life Long Learning	PSO-1	PSO-2
CO-1	3	-	-	-	-	-	-		
CO-2	3	1	-	-	-	-	-		
CO-3	3	1	1	-	-	-	1		
CO-4	3	1	-	-	-	-	-		
CO-5	3	2	1	1	-	-	1		

Legend: High (3), Medium (2), Low (1) and No mapping (-)

* PSOs will be developed by the respective program coordinators at the institute level. As per the latest NBA guidelines, formulating PSOs is optional

G) Teaching & Learning Scheme:

Course Code	Course Title	Scheme of Study (Hours/Week)					
		Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+ SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
		L	T				
2400101	Basic Engineering Mathematics	02	01	-	02	05	04

Legend:

CI: Classroom Instruction (Includes different instructional/ implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem based learning etc. to deliver theoretical concepts)

LI: Laboratory Instruction (Includes experiments/ practical performances / problem-based experiences in laboratory, workshop, field or other locations using different instructional/ Implementation strategies)

Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.

TW: Term Work (includes assignments, seminars, micro projects, industrial visits, any other student activities etc.)

SL: Self Learning, MOOCs, Spoken Tutorials, online educational resources etc.

C: Credits= (1xCI hours) + (0.5xLI hours) + (0.5xNotional hours)

Note: TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of teacher to ensure outcome of learning.

H) Assessment Scheme:

Course Code	Course Title	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
		Theory Assessment (TA)		Term Work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
		Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2400101	Basic Engineering Mathematics	30	70	20	30	-	-	150

Legend:

PTA: Progressive Theory Assessment in the classroom (includes class test, mid-term test, and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro-projects, industrial visits, self-learning, any other student activities, etc.)

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignments, micro-projects, seminars, and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/ presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria for internal as well as external assessment may vary as per the requirement of the respective course. For valid and reliable assessment, the internal faculty should prepare a checklist & rubrics for these activities.

I) Course Curriculum Detailing: This course curriculum detailing depicts learning outcomes at course level and session level and their attainment by the students through Classroom Instruction (CI), Laboratory Instruction (LI), Term Work (TW), and Self Learning (SL). Students are expected to demonstrate the attainment of Theory Session Outcomes (TSOs) and Lab Session Outcomes (LSOs) leading to the attainment of Course Outcomes (COs) upon the completion of the course. While curriculum detailing, NEP 2020-related reforms like Green skills, Sustainability, Multidisciplinary aspects, Society connect, Indian Knowledge System (IKS), and others must be integrated appropriately.

J) Theory Session Outcomes (TSOs) and Units: T2400101

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO 1a.</i> Find the solution of a system of equations in three unknowns by applying Cramer's rule. <i>TSO 1b.</i> Solve simple given problems based on the Algebra of matrices. <i>TSO 1c.</i> Find the inverse of the matrix by applying the concept of Adjoint of the matrix. <i>TSO 1d.</i> Find a solution of simultaneous equations in three variables using the concept of the Matrix Inversion method. <i>TSO 1e.</i> Solve problems based on the sum, and subtraction of Vectors. <i>TSO 1f.</i> Solve simple problems related to Scalar and Vector product of vectors. <i>TSO 1g.</i> Solve simultaneous equations by using concepts given in Ancient Indian Mathematics. (IKS)	Unit-1.0 Algebra Determinant 1.1 Concept and properties of determinant. 1.2 Solutions of simultaneous equations in three Unknowns by Cramer's rule. Matrices 1.3 Algebra of matrices (Addition, Subtraction, Multiplication by Scalar, and Multiplication of Two matrices). 1.4 Transpose, Adjoint and Inverse of Matrix. 1.5 Solutions of simultaneous equations of a Matrix of order 3 x3 by Inversion method. Vectors 1.6 Position vector. 1.7 Algebra of Vectors (Addition, Subtraction, Scalar Multiplication with vector). 1.8 Scalar product. 1.9 Vector product. 1.10 Algebra in Indian Knowledge System: Solution of simultaneous equations (Indian Mathematics). (IKS)	CO1
<i>TSO 2a.</i> Define the concept of a function and its types. <i>TSO 2b.</i> Solve simple problems based on Domain and range of function. <i>TSO 2c.</i> Evaluate problems of limit function based on Indeterminate form. <i>TSO 2d.</i> Check the continuity of a function at a point. <i>TSO 2e.</i> Find the differentiation of some simple functions (sinx, cosx, tanx, and e^x) by the first principle. <i>TSO 2f.</i> Calculate the derivative of given Algebraic, trigonometric, and exponential functions. <i>TSO 2g.</i> Find the derivative of the given two functions' sum, product, and quotient. <i>TSO 2h.</i> Find the differentiation of given composite functions by applying the concept of the Chain rule. <i>TSO 2i.</i> Find the derivative of Logarithmic, Implicit, and Parametric functions. <i>TSO 2j.</i> Familiar with the concept of calculus given in Indian Mathematics. (IKS)	Unit-2.0 Differential Calculus Function and Limit 2.1 Concept of function. 2.2 Different type of functions. 2.3 Domain and Range of Function. 2.4 Concept of Limits and its evaluation. Continuity 2.5 Concept of continuity with simple problems. Differentiation 2.6 Differentiation by First Principle. 2.7 Differentiation of Algebraic, trigonometric, Exponential, and Logarithmic functions. 2.8 Differentiation of sum, product, and quotient of two functions. 2.9 Differentiation of composite functions by Chain Rule. 2.10 Logarithmic differentiation. 2.11 Implicit differentiation. 2.12 Differentiation of Parametric Functions. 2.13 Calculus in Indian Knowledge System: The Discovery of Calculus by Indian Astronomers. (Indian Mathematics). (IKS)	CO2
<i>TSO 3a.</i> Find the second-order derivative of given simple functions. <i>TSO 3b.</i> Solve simple problems based on Rolle's Theorem and Mean Value Theorem.	Unit-3.0 Application of Differential Calculus 3.1 Successive differentiation up to second order. 3.2 Rolle's Theorem and Mean Value Theorem (without proof) with examples. 3.3 Rate of change of quantities.	CO3

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO 3c.</i> Apply the concept of Rate of change to solve simple problems related to velocity, and acceleration. <i>TSO 3d.</i> Apply rules of derivative to solve given applied problems related to tangent and normal. <i>TSO 3e.</i> Apply rules of derivative to solve applied problems based on Maxima-Minima and Radius of curvature.	3.4 Equation of Tangent and Normal. 3.5 Maxima and Minima. 3.6 Radius of curvature.	
<i>TSO 4a.</i> Calculate the angle between the given two lines and also find the slope. <i>TSO 4b.</i> Formulate an equation of straight lines of different forms. <i>TSO 4c.</i> Find the perpendicular distance of a straight line from a given point and the perpendicular distance between two parallel lines. <i>TSO 4d.</i> Use the geometry given in Sulbasutras to solve the given problems. <i>TSO 4e.</i> Solve simple problems related to Circles and Parabola for engineering applications. <i>TSO 4f.</i> Solve given simple problems related to Ellipse for engineering applications.	Unit-4.0 Co-ordinate Geometry Co-ordinate systems 4.1 Introduction of Co-ordinate Systems. Straight lines 4.2 Slope of a line, the angle between two lines. Various forms of Straight Lines 4.3 Point-slope form, Two-point form, Slope intercept form, Intercept form, Normal form, General form. 4.4 Perpendicular distance of a line from a point, perpendicular distance between two parallel lines. 4.5 Geometry in Sulbasutras in Indian Knowledge System (construction of the square, circling the square). (Indian Mathematics). Conic Section 4.6 Introduction of Conic-Section. 4.7 Equation of Circle in standard form. 4.8 Standard equation of parabola, ellipse, and hyperbola.	CO4
<i>TSO 5a.</i> Compute the probability of given simple problems based on the Addition and Multiplication theorem. <i>TSO 5b.</i> Evaluate the Mean, Median, and Mode of the given data for engineering applications. <i>TSO 5c.</i> Calculate the Range, Variance, and standard deviation of given data for engineering applications. <i>TSO 5d.</i> Calculate the Coefficient of variance of given data for engineering applications.	Unit-5.0 Probability and Statistics Probability 5.1 Concept of Probability. 5.2 Addition and multiplication theorems of Probability. The measure of Central Tendency 5.3 Mean, Median, Mode. Measure of Dispersion 5.4 Range, Variance, Standard Deviation. 5.5 Coefficient of Variation.	CO5

Note: One major TSO may require more than one Theory session/Period.

K) Suggested Tutorials and Outcomes:

Outcomes	S. No.	Tutorials Titles	Relevant COs Number(s)
1.1 Determine the value of the determinant by using available open-source software. 1.2 Determine the inverse of a non-singular matrix by using open-source software. 1.3 Apply the Matrix Inversion method to determine currents through various branches of given electrical networks. 1.4 Determine the resultant force applied at a particle using properties of vector for a given engineering problem.	1.	- Value of determinant of order 3, 4, and higher using opensource software. - Inverse of the non-singular matrix using open-source software. - Calculation of current in electrical networks by Matrix Inversion method. - Geometrical interpretation of operations of vector algebra.	CO1
2.1 Geometrically represent the domain and range of the given Modulus function, Signum function, and Floor function. 2.2 Verify geometrically the continuity of a given function at a point. 2.3 Determine the concavity and convexity of a given continuous function for a given engineering application. 2.4 Find the acceleration of the given moving body at a time t.	2.	- Geometrical interpretation of domain and range of a function. - Geometrical interpretation of limit and continuity. - Branch-specific engineering application of derivative. - Branch-specific engineering application of derivative of a parametric function.	CO2
3.1 Determine the maximum height of a projectile trajectory using Roll's theorem. 3.2 Use Lagrange's Mean Value theorem to find the point at which the slope of the tangent becomes equal to the slope of the secant through its endpoints. 3.3 Use the concept of derivative to find the slope of a bending curve for a given engineering problem. 3.4 Use the concept of tangent and normal to solve the given problem of Engineering Drawing. 3.5 Use the concepts of Maxima and Minima to obtain optimum value for a given engineering problem. 3.6 Use the concept of the radius of curvature to solve a given branch-specific engineering problem.	3.	- Geometrical Interpretation of Rolle's Theorem. - Geometrical Interpretation of Lagrange's Mean Value theorem. - Branch-specific engineering application of rate of change of quantities. - Branch-specific engineering applications of tangent and normal. - Branch-specific engineering applications of maxima and minima. - Engineering applications of Radius of curvature.	CO3
4.1 Apply the concept of Gradient to draw graphs in engineering drawing. 4.2 Use the given form of a straight line to calculate the speed, distance, and time of a moving object. 4.3 Use the concept of Ellipse to prepare a Model of the path of the Planet and its foci.	4.	- Geometrical interpretation of Gradient. - Geometrical Interpretation of lines in various forms. - Geometrical interpretation of the perpendicular distance of a line. - Geometrical representation of conic-section.	CO4
5.1 Use the concept of probability to solve given problems based on Board and playing cards. 5.2 Calculate the Standard Deviation for Concrete with the given data.	5.	- Applications of Probability and related theorems. - Applications of Mean, Median, and Mode for applied problems.	CO5

- L) **Suggested Term Work and Self-Learning: S2400101** Some sample suggested assignments, micro-projects, and other activities are mentioned here for reference.
- a. **Assignments:** Questions/Problems/Numerical/Exercises to be provided by the course teacher in line with the targeted COs.
1. Solve the simultaneous system of equations in two variables by Matrix Inversion Method. Write down a Mathematical program using any open-source software to verify the result.
 2. A rigid body is subjected to multiple forces acting at different points. Apply vector technique to calculate the net moment or torque acting on the body. Discuss the equilibrium condition and the significance of the moment in terms of structural integrity and mechanical system using open-source software.
 3. Represent the Graph of the Trigonometric function and logarithmic function on GeoGebra. Interpret the nature of the graph and make a pdf file.
 4. Find the derivative of $y = x^{\sin x}$ and visualize the graph of the function and its derivative using any open-source software geometrically.
 5. A window in the form of a rectangle surmounted by a semi-circular opening. The total perimeter of the window to admit maximum light through the whole opening. Prepare a model using the concept of Maxima and Minima for the above problem and verify the result.
 6. Find the curvature of $x = 4\cos t$ and $y = 3\sin t$, at what point on this ellipse does the curvature have the greatest and least values? What are the magnitudes? Visualize the result graphically using any open-source software.
 7. When a double-sided right circular cone is intersected by a plane, different types of conic sections are generated. Represent all these conic sections on GeoGebra and write down their equation.
 8. Explain how parabolic reflectors are used in engineering applications such as Satellite Dish Antennas or headlights.
 9. By Collecting the Data of the Last 5 IPL series, Calculate the probability of winning a match by any two teams.
 10. Collect the Data of Marks obtained by your class in 1st class test. Compute the Mean, Median, Mode, and variance of the data and interpret the result.
- b. **Micro Projects:**
1. Prepare charts displaying the properties of determinants and Matrices.
 2. Prepare a chart for the use of Vector algebra to solve problems of the rate of change of the mass of a fluid flow.
 3. Draw the graph of functions like x^2 , $\sin x$, $\cos x$, $\tan x$, and e^x etc analytically on graph paper and verify using suitable open-source software like Sage Math, Math3d, GeoGebra, Wolfram Alpha, and Dplot and prepare a pdf file.
 4. Collect at least 10 engineering applications for each Limits, Continuity, and Differentiability and prepare a PDF file.
 5. Prepare a chart consisting of 8-10 engineering-related functions whose derivative does not exist.
 6. Prepare a model showing the application of Rolle's Theorem to determine the projectile trajectories of maximum height.
 7. Prepare a chart consisting of any 10 applications of the Mean value theorem related to real-world problems.
 8. Model to maximize the volume of a box made of a rectangular tin sheet by cutting off squares of the same size from each corner and folding them up. Also, design models for at least 5 similar situations and prepare a soft file with animation.
 9. Prepare models using the concept of tangent and normal while bending of roads in case of sliding of a vehicle.

10. Prepare models using the concept of the radius of curvature while bending of railway track.
11. Make a short video of duration 5-7 minutes for the use of Derivative to calculate the profit and loss in business using graphs.
12. Download 5-7 videos based on applications of Derivative to check the temperature variation, find the range of magnitudes of the earthquake, etc. Watch them and write a report to detail the mathematical steps involved.
13. Prepare the Charts of formulae showing different forms of straight lines for engineering applications.
14. Draw the graph for the standard equations of Circle, Parabola, Ellipse, and Hyperbola on the Chart paper using any open-source software and make a file.
15. Prepare the Charts consisting tree diagram to find the probability of a given event.
16. Collect the data of World of Work and find the mean, mean deviation, and standard deviation for that data using any open-source software of Statistics and make a soft copy.
17. Download 5-7 videos based on applications of probability for the weather forecast, watch them, and write a report to detail the mathematical steps involved.

c. Other Activities:

1. Seminar Topics:

- Applications of Integral calculus in control systems, dynamics, and vibrations.
- Applications of determinants and matrices in graphic design to make digital images.
- Application of determinants and matrices for calculating the battery power outputs.
- Application of Vector algebra in engineering mechanics.
- Application of limit and continuity to measure the strength of the magnetic field and electric field.
- Applications of Derivatives for engineering & technology.
- Application of radius of curvature for Engineering and Science.
- Applications of Derivatives in the economy to compute the level of output at which the total revenue is the highest, the profit is the highest, and (or) the lowest, etc.
- Applications of Coordinate geometry to design of athletic tracks, recreational parks, building plans, roundabouts, Ferris wheels.
- Application of ellipses to be used to orbits of planets, satellites, moons comets, etc.
- Probability and statistics: Civil engineering, estimation of model uncertainties, identification of probability distribution.

2. Visits: Visiting the following places would provide students an opportunity to see the application of various branches of mathematics in different fields. This will also help students to comprehend the career opportunities available in the field of mathematics.

- Visit to a mathematics museum.
- Visit a mathematics research institute.
- Visit to a mathematics laboratory.
- Visit to a Data Science Center.
- Visit the mathematics department of a college or university.
- Visit a mathematics software company.
- Visit to a Cryptography Company.
- Visit to a Space Agency.
- Visit to a Game Studio.
- Visit to a mathematics library.
- Attend Mathematical conferences on real-world problem-solving.
- Participation in mathematics competitions.

3. Self-Learning Topics:

- Participate in MOOCs based Course on Matrix offered by Foreign University: Methods and Applications.
- Participate in an MOOCs-based Course on Differential Calculus: Methods and Applications.
- Participate in MOOC-based Courses on Probability and its Engineering applications.
- Participate in MOOC-based Courses on Statistics and its Engineering applications.
- Watching videos on applications of coordinate geometry to Real-world problems.

M) Suggested Course Evaluation Matrix: The course teacher has to decide and use the appropriate assessment strategy and its weightage in theory, laboratory, and Term Work for ensuring CO attainment. The response/performance of each student in each of these designed activities is to be used to calculate **CO attainment**.

COs	Course Evaluation Matrix						
	Theory Assessment (TA)**		Term Work Assessment (TWA)			Lab Assessment (LA)#	
	Progressive Theory Assessment (PTA) Class/Mid Sem Test	End Theory Assessment (ETA)	Term Work & Self Learning Assessment			Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)
			Assignments	Micro Projects	Other Activities*		
CO-1	20%	20%	15%	15%	25%	-	-
CO-2	15%	20%	20%	20%	15%	-	-
CO-3	20%	15%	15%	15%	10%	-	-
CO-4	20%	20%	25%	25%	25%	-	-
CO-5	25%	25%	25%	25%	25%	-	-
Total Marks	30	70	20	20	10	-	-
			50				

Legend:

*: Other Activities include self-learning, seminars, visits, surveys, product development, software development, etc.

** : Mentioned under point- (N)

: Mentioned under point-(O)

Note:

- The percentages given are approximate
- In the case of Micro Projects and End Laboratory Assessment (ELA), the achieved marks will be equally divided among all those COs mapped with total experiments.
- For CO attainment calculation indirect assessment tools like course exit survey need to be used which comprises questions related to the achievement of each CO.

- N) Suggested Specification Table for End Semester Theory Assessment:** The specification table represents the reflection of sample representation of the assessment of the cognitive domain of the full course.

Unit Title and Number	Total Classroom Instruction (CI) Hours	Relevant COs Number(s)	Total Marks	ETA (Marks)		
				Remember (R)	Understanding (U)	Application & above (A)
Unit-1.0 Algebra	8	CO1	12	4	4	4
Unit-2.0 Differential Calculus	10	CO2	14	4	8	2
Unit-3.0 Application of Differential Calculus	8	CO3	12	4	4	4
Unit-4.0 Co-ordinate Geometry	10	CO4	14	4	6	4
Unit-5.0 Probability and Statistics	12	CO5	18	4	6	8
Total	48	-	70	20	28	22

Note: A similar table can also be used to design class/mid-term/ internal question papers for progressive assessment.

- O) Suggested Assessment Table for Laboratory (Practical): (Not Applicable)**

- P) Suggested Instructional/ Implementation Strategies:** Different Instructional/ Implementation Strategies may be appropriately selected, as per the requirement of the content/outcome. Some of them are Improved Lectures, Tutorial, Case Methods, Group Discussions, Industrial visits, Industrial Training, Field Trips, Portfolios, Learning, Role Play, Live Demonstrations in Classrooms, Labs, Field Information, and Communications Technology (ICT) Based Teaching Learning, Blended or flipped mode, Brainstorming, Expert Session, Video Clippings, Use of Open Educational Resources (OER), MOOCs etc.

- Q) List of Major Laboratory Equipment, Tools and Software:**

S. No.	Name of Equipment, Tools and Software	Broad Specifications	Relevant Experiment/Practical Number
1.	High-end computers	Processor Intel Core i7 with Compilers and Programming Languages; RAM 32 GB, DDR3/DDR4, HDD 500 GB, OS Windows 10.	All
2.	Software	Scientific Calculators, Graphing Calculator, SCILAB, Graph Eq ^{2.13} , Microsoft Mathematics, GeoGebra, Math3D	1,2,3,4,5
3.	Printer	High-Speed Duplex Printer	
4.	Scanner	Handheld 3D scanner, Accuracy up to 0.1 mm, Resolution up to 0.2 mm, Wireless technology with an inbuilt touch screen and battery, Extended field of view for capturing both large and small objects.	

- R) Suggested Learning Resources:**

(a) Books:

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
1.	Elementary Engineering Mathematics	B. S. Grewal	Khanna Publishers, 15th Edition. ISBN: 978-81-7409-257-1
2.	Engineering Mathematics (Third edition)	Croft, Anthony	Pearson Education, New Delhi, 2014. ISBN 978-81-317-2605-1
3.	Calculus and Its Applications	Marvin L. Bittinger David J. Ellenbogen Scott A. Sargent	Addison-Wesley 10th Edition ISBN-13: 978-0-321-69433-1
4.	Calculus and Analytic Geometry	G. B. Thomas, R. L. Finney	Addison Wesley, 9th Edition, 1995. ISBN 978-8174906168
5.	Understanding Engineering Mathematics	John Bird	Routledge; First Edition ISBN 978-0415662840
6.	Advanced Engineering Mathematics	Krezig, Ervin	Wiley Publ., New Delhi, 2014, ISBN: 978-0-470-45836-5
7.	Indian Mathematics Engaging with the World from Ancient to Modern Times	George Gheverghese Joseph	World Scientific Publishing Europe Ltd. 57 ISBN 978-17-86340-61-0
8.	A Modern Introduction to Ancient Indian Mathematics	T.S. Bhanumurthy	New Age International Private Limited, 1 January 2008 ISBN- 10. 812242600X, ISBN- 13. 978-8122426007
9.	Mathematics-I	Deepak Singh	Khanna Book Publishing Co. (P) Ltd. ISBN: 978-93-91505-42-4
10.	Mathematics-II	Garima Singh	Khanna Book Publishing Co. (P) Ltd. ISBN: 978-93-91505-52-3
11.	Consider Dimension and Replace Pi	M.P. Trivedi and P.Y. Trivedi	Notion Press; 1 st edition (2018), ISBN: 978-1644291795
12.	Sansar Ke Mahan Ganitagya	Gunakar Muley	First Edition, Rajkamal Prakashan, ISBN-10. 8126703571, ISBN-13. 978- 8126703579.

(b) Online Educational Resources:

1. <https://ocw.mit.edu/>
2. <https://tutorial.math.lamar.edu/>
3. <https://www.khanacademy.org/>
4. <https://www.feynmanlectures.caltech.edu/>
5. <https://www.wolframalpha.com/>
6. <https://www.dplot.com/>
7. <https://www.geogebra.org/>
8. <https://www.easycalculation.com/>
9. <https://www.scilab.org/>
10. <https://www.desmos.com/>
11. <https://nptel.ac.in/>
12. <https://swayam.gov.in/>
13. <https://ndl.iitkgp.ac.in/>
14. <https://parakh.aicte-india.org/>
15. <https://ekumbh.aicte-india.org/>
16. <https://learnengg.com/LE/Index>
17. <https://ncert.nic.in/textbook.php>
18. [https://nios.ac.in/online-course-material/sr-secondary-courses/mathematics-\(311\).aspx](https://nios.ac.in/online-course-material/sr-secondary-courses/mathematics-(311).aspx)

Note: Teachers are requested to check the Creative Commons license status/ financial implications of the suggested, online educational resources before use by the students.

(c) Others:

1. Online Mathematics Courses.
2. Mathematics Communities and Forums.
3. Mathematics Journals.
4. Mathematics Podcast.
5. Mathematics Tutorials.
6. Mathematics Quizzes.
7. Mathematics Animation.
8. Mathematics Simulations.
9. Mathematics Games.
10. Mathematics Puzzles.
11. Mathematics Brain Teasers.
12. Mathematics Apps.
13. Mathematics Blog.
14. Mathematics Challenges.

- A) **Course Code** : 2400103A(T2400103A/P2400103A/S2400103A)
- B) **Course Title** : Applied Chemistry- A (ME, ME (Auto), CE, MIE, AE, CHE, FTS, CRE)
- C) **Pre- requisite Course(s)** :
- D) **Rationale** :

Students pursuing diplomas in engineering fields like mechanical, automobile, civil, mining, chemical, ceramic, agricultural, fire technology and safety need to study applied chemistry as a prerequisite course. After completion of this course student will have a deep understanding of chemical concepts, their uses, and how they relate to engineering field. Diploma engineers deals with various concept of chemistry to be approved in diverse technical and engineering field. Ever increasing use of materials like metals, alloys and fuel and lubricants will compel engineers to acquire essential applied chemistry knowledge to select engineering material, which will be economical and eco-friendly. Through this course, they will be able to understand structural arrangement of fundamental particles, atoms and molecules. The knowledge of chemical bonding will help the engineers and scientist to design new engineering materials and form chemical compounds with desirable properties. The study of concepts like water treatment and analysis, fuels and combustions and electrochemistry have constantly proved the importance of applied chemistry.

- E) **Course Outcomes (COs):** After the completion of the course, teachers are expected to ensure the accomplishment of following course outcomes by the learners. For this, the learners are expected to perform various activities related to three learning domains (Cognitive, Psychomotor and Affective) in classroom/ laboratory/ workshop/ field/ industry.

After completion of the course, the students will be able to-

- CO-1** Solve various engineering problems applying the basic concepts of atomic structure, chemical bonding, and solutions.
- CO-2** Use relevant water treatment techniques to solve domestic and industrial problems.
- CO-3** Solve engineering problems using concepts of engineering materials and properties.
- CO-4** Use relevant fuels and lubricants for domestic and industrial applications.
- CO-5** Solve engineering problems using the concepts of electrochemistry and corrosion.

- F) **Suggested Course Articulation Matrix (CAM):**

Course Outcomes (COs)	Programme Outcomes(POs)							Programme Specific Outcomes* (PSOs)	
	PO-1 Basic and Discipline Specific Knowledge	PO-2 Problem Analysis	PO-3 Design/ Development of Solutions	PO-4 Engineering Tools	PO-5 Engineering Practices for Society, Sustainability and Environment	PO-6 Project Management	PO-7 Life Long Learning	PSO-1	PSO-2
CO-1	3	2	1	-	-	-	1		
CO-2	3	3	2	2	2	1	1		
CO-3	3	2	1	2	1	1	1		
CO-4	3	1	1	1	1	1	1		
CO-5	3	2	1	1	-	1	2		

Legend: High (3), Medium (2), Low (1) and No mapping (-)

* PSOs will be developed by respective programme coordinator at institute level. As per latest NBA guidelines, formulating PSOs is optional

G) Teaching & Learning Scheme:

Course Code	Course Title	Scheme of Study (Hours/Week)					
		Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+ SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
		L	T				
2400103A	Applied Chemistry- A	03	-	04	02	09	06

Legend:

CI: Classroom Instruction (Includes different instructional/implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem based learning etc. to deliver theoretical concepts)

LI: Laboratory Instruction (Includes experiments/practical performances /problem-based experiences in laboratory, workshop, field or other locations using different instructional/Implementation strategies)

Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.

TW: Term Work (includes assignments, seminars, micro projects, industrial visits, any other student activities etc.)

SL: Self Learning, MOOCs, spoken tutorials, online educational resources etc.

C: Credits = (1 x CI hours) + (0.5 x LI hours) + (0.5 x Notional hours)

Note: TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of teacher to ensure outcome of learning.

H) Assessment Scheme:

Course Code	Course Title	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
		Theory Assessment (TA)		Term Work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
		Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2400103A	Applied Chemistry- A	30	70	20	30	20	30	200

Legend:

PTA: Progressive Theory Assessment in class room (includes class test, mid-term test and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro projects, industrial visits, self-learning, any other student activities etc.)

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignment, micro project, seminar and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/ presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria of internal as well as external assessment may vary as per the requirement of respective course. For valid and reliable assessment, the internal faculty should prepare checklist & rubrics for these activities.

- I) Course Curriculum Detailing:** This course curriculum detailing depicts learning outcomes at course level and session level and their attainment by the students through Classroom Instruction (CI), Laboratory Instruction (LI), Term Work (TW) and Self Learning (SL). Students are expected to demonstrate the attainment of Theory Session Outcomes (TSOs) and Lab Session Outcomes (LSOs) leading to attainment of Course Outcomes (COs) upon the completion of the course. While curriculum detailing, NEP 2020 related reforms like Green skills, Sustainability, Multidisciplinary aspects, Society connect, Indian Knowledge System (IKS) and others must be integrated appropriately.

J) Theory Session Outcomes (TSOs) and Units: T2400103A

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO-1a</i> Describe the three subatomic particles in an atom. <i>TSO-1b</i> Conclude Rutherford model of atom. <i>TSO-1c</i> Apply the different atomic theories and principles for structural illustration. <i>TSO-1d</i> Calculate uncertainty in position and momentum. <i>TSO-1e</i> Draw the shapes of s, p and d orbitals. <i>TSO-1f</i> Write the electronic configuration of different elements. <i>TSO-1g</i> Differentiate between ionic, covalent, and coordinate compounds based on the type of chemical bonding. <i>TSO-1h</i> Explain the unique behavior of water. <i>TSO-1i</i> Prepare the solution of given concentration.	Unit-1.0 Atomic Structure and Chemical Bonding and Solutions: 1.1 Atoms and its fundamental particles. 1.2 Rutherford Model of Atom. 1.3 Bohr's Theory, Hydrogen spectrum explanation based on Bohr's Model of Atom. 1.4 Wave Mechanical model of atom, de Broglie relationship, Heisenberg Uncertainty Principle 1.5 Quantum Numbers, Shapes of Atomic Orbitals. 1.6 Pauli's Exclusion Principle, Hund's Rule of Maximum Multiplicity, Aufbau Principle, Electronic Configuration (till atomic number 30). 1.7 Concept of Chemical bonding - Cause of chemical bonding, Types of Bonds: Ionic Bond (NaCl, CaCl₂, MgO), Covalent Bond, Polar and Nonpolar Covalent Bonds (H₂, F₂, HF, HCl) & Co-ordinate Bond (CO, NH₄⁺, O₃, H₂SO₄), 1.8 Dipole Moment (NH₃, NF₃), Hydrogen bonding. 1.9 Solution- (solute, solvent) and their Strength- Molarity, Normality, Molality. 1.10 Indian Chemistry: -Philosophy of atom by Acharya Kanad. (IKS)	CO1
<i>TSO-2a</i> Classify hard and soft water based on their properties. <i>TSO-2b</i> List the impurities responsible for hardness. <i>TSO-2c</i> Calculate the hardness of water. <i>TSO-2d</i> Determine the hardness by EDTA method. <i>TSO-2e</i> Apply different water softening techniques to soften the hard water. <i>TSO-2f</i> Calculate the amount of lime and soda required for removal of hardness. <i>TSO-2g</i> Differentiate between BOD and COD. <i>TSO-2h</i> Use the Indian standard specification of drinking water.	Unit-2.0 Water 2.1 Introduction, Sources of Water. Hardness of Water- Temporary & Permanent hardness. 2.2 Degree of Hardness (In terms of CaCO₃ equivalent), Unit of Hardness, Quantitative Measurement of Water Hardness by EDTA method. 2.3 Municipal supply of Water, Treatment of water, Water Softening Technique-Soda Lime Process, Zeolites method and ion exchange method. 2.4 Water Quality Index - Biological Oxygen Demand, Chemical Oxygen Demand, Determination of Dissolved Oxygen 2.5 Indian standard specification of drinking water.	CO2
<i>TSO-3a</i> List ores of metals. <i>TSO-3b</i> Describe ore, gangue, matrix. <i>TSO-3c</i> Select Appropriate metallurgical processes for concentration, extraction, and purification of given ore. <i>TSO-3d</i> Describe alloy with examples. <i>TSO-3e</i> Write the constituent of given alloy. <i>TSO-3f</i> Write the composition properties and uses of ferrous and non-ferrous alloys. <i>TSO-3g</i> Distinguish between homopolymer and copolymer. <i>TSO-3h</i> Write the monomers of given polymers. <i>TSO-3i</i> Explain vulcanization process. <i>TSO-3j</i> Explain cement & its manufacture.	Unit-3.0 Engineering materials 3.1 Natural Occurrence of Metals- Minerals, ores. 3.2 Metallurgy - General principles of Metallurgy, Gangue, Flux and Slag, Steps involved in metallurgy. 3.3 Extraction of Aluminium, Iron and Copper from their important ores along with reactions, Properties and uses. 3.4 Alloys – Definition, Purpose of alloying, Ferrous and Non-Ferrous Alloy with suitable examples, Composition, Properties, and their applications. 3.5 Ancient Indian Metallurgy (IKS) 3.6 Polymers-Homopolymers and Copolymers, Natural polymers and synthetic polymers,	CO3

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO-3k</i> Differentiate among the different engineering materials based on their chemical composition and composition-based applications.	Addition and Condensation polymerization, Thermoplastic and Thermosetting plastic. 3.7 Monomers, applications, and synthesis of Polythene, PVC, Orlon, Terylene, Nylon 66, Nylon 6, Bakelite. 3.8 Natural Rubber and its vulcanization, advantages of vulcanized rubber. 3.9 Cement, Average composition of Portland cement, Raw material for manufacture of cement, Setting of Cement.	
<i>TSO-4a</i> Classify fuels. <i>TSO-4b</i> Describe HCV and LCV. <i>TSO-4c</i> Explain knocking, octane number and cetane number. <i>TSO-4d</i> Use different gaseous fuels based on their composition, calorific value, and other properties. <i>TSO-4e</i> Explain uses of NPK fertilizers. <i>TSO-4f</i> Select relevant lubricant based on their composition, calorific value, and other properties. <i>TSO-4g</i> Determine viscosity, flash, and fire point of given lubricant for its specific use. <i>TSO-4h</i> Explain Flash, Fire, Cloud & Pour point.	Unit-4.0 Chemistry of Fuel and Lubricants 4.1 Fuels, Characteristics of an Ideal Fuel. 4.2 Classification of Fuel- Solid, liquid and gas fuel, Calorific Values (HCV and LCV), 4.3 Petroleum and its fractional distillation. 4.4 Cracking, knocking, Fuel Rating (Octane Number, Cetane Number). 4.5 Composition, uses, advantages and disadvantages of LPG, CNG and Biogas. 4.6 Manures, NPK fertilizers (preparation and uses). 4.7 Fire Extinguishers and their types. 4.8 Lubricants- Classification of Lubricants with examples, Functions and Properties of Good Lubricant. 4.9 Viscosity & Viscosity Index. Flash point. Fire point, Cloud & Pour point	CO4
<i>TSO-5a.</i> Describe Electrolyte and Nonelectrolyte. <i>TSO-5b.</i> Describe Metallic and electrolytic conduction. <i>TSO-5c.</i> Explain the faraday law of electrolysis. <i>TSO-5d.</i> Calculate the mass of metal deposited after passing a certain amount of current. <i>TSO-5e.</i> Calculate the emf at different temperature, pressure, and molar concentration. <i>TSO-5f.</i> Predict the feasibility of a cell. <i>TSO-5g.</i> Explain the working of a cell. <i>TSO-5h.</i> Describe corrosion. <i>TSO-5i.</i> Explain the different methods to prevent corrosion.	Unit-5.0 Electrochemistry 5.1. Introduction, Electrolyte and Nonelectrolyte, Electrolytic and Metallic Conduction, Factors affecting Electrolytic Conductance. 5.2. Molar Conductivity and Equivalent Conductivity. Variation of Molar Conductivity, Kohlrausch's law. 5.3. Faraday's Laws of Electrolysis. 5.4. Galvanic Cell, Electrode Potential, Measurement of Electrode Potential SHE (Standard Hydrogen electrode), EMF, Electrochemical Series, Nernst Equation for Electrode Potential. 5.5. Batteries, Primary Cells-Dry cell, Secondary cell -Lead storage battery, Fuel cells. 5.6. Corrosion, their types (Dry & Wet corrosion) and prevention.	CO5

Note: One major TSO may require more than one Theory session/Period.

K) Suggested Laboratory (Practical) Session Outcomes (LSOs) and List of Practical: P2400103A

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
LSO-1.1. Calculate amount of oxalic acid required. LSO-1.2. Prepare N/10 oxalic acid solution.	1.	Preparation of 250 ml of N/10 Oxalic acid Solution	CO1
LSO-2.1. Calculate amount of Sodium Carbonate required. LSO-2.2. Prepare N/10 Sodium Carbonate Solution.	2.	Preparation of 250ml of N/10 Sodium Carbonate Solution.	CO1
LSO 3.1. Perform acid base titration. LSO 3.2. Prepare oxalic acid solution	3.	Determination of strength of Sodium Hydroxide solution by titrating against Oxalic Acid Solution	CO1
LSO 4.1. Perform Complexometric titration. LSO 4.2. Standardize EDTA solution.	4.	Determination of the total hardness of tap water by EDTA method	CO2
LSO 5.1. Calculate % of moisture	5.	Estimation of moisture content in given coal sample gravimetrically.	CO4
LSO-6.1. Perform double displacement reaction. LSO-6.2. Test the presence of sulphate.	6.	Preparation of Barium Sulphate from Barium Chloride.	CO2
LSO-7.1. Use viscometer. LSO-7.2. Calculate viscosity using the drop number method.	7.	Determination of viscosity of liquid Using Ostwald Viscometer.	CO4
LSO-8.1. Construct Daniel cell. LSO-8.2. Compare the effect of dilution of electrolytes on the emf of a Daniel cell.	8.	Comparison of the effect of dilution of electrolytes on the emf of a Daniel cell.	CO5
LSO 9.1. Perform acid base titration using pH meter.	9.	Determination of pH of given solution by pH meter.	CO2
LSO-10.1. Carry out Polymerization. LSO-10.2. Set the environment for carrying out polymerization.	10.	Preparation of Phenol Formaldehyde Resin (Bakelite).	CO3
LSO-11.1. Perform iodometry titration. LSO-11.2. Use of starch as indicator.	11.	Determination of dissolved Oxygen in given sample of Water.	CO2
LSO-12.1. Calculate pH.	12.	Determination of pH of soil using baking soda and vinegar.	CO2

L) Suggested Term Work and Self Learning: S2400103A Some sample suggested assignments, micro project and other activities are mentioned here for reference.

a. Assignments: Questions/Problems/Numerical/Exercises to be provided by the course teacher in line with the targeted Cos such as

1. Write electronic structure of given atoms.
2. Compare the wavelengths of different macroscopic and microscopic particles moving with same velocity.
3. Prepare a model to find the soap lather forming capacity of tap water on addition of lime.
4. Prepare chart showing different industrial application of metal and relate it with required property or properties using internet.
5. Explain the working principle of TEL as antiknock.
6. Prepare chart showing different types of liquid fuels with their calorific values and uses.
7. Prepare a comparative chart of commercially available lubricants based on mechanism of lubrication.
8. Compare the EMF of Zinc - Copper cell with different cathodic concentration and predict out of low and high cathodic concentration, which increases EMF?
9. Prove the statement mathematically. "It is impossible to determine the position and momentum simultaneously with accuracy."

b. Micro Projects:

1. Form three groups of students in the class. Consider a hypothetical situation of exchanging/ sharing/giving of different items/belongings and demonstrate the type of ionic, covalent, and co-ordinate bonding amongst the students in a simulated situation. Present your findings.
2. Prepare a model of electronic configurations for different atoms ($Z=30$)
3. Prepare a model to demonstrate the application of electrolytic cells.
4. Collect three metallic strips of Al, Cu, Fe, strips, Place them in different acidic and alkaline solutions of the same concentration. Observe and record the loss in weight of metals due to acidic and alkaline environments. Discuss the findings with your teacher and colleagues.
5. Classify the surrounding corrosion into dry corrosion and wet corrosion.
6. Collect different samples of utensils reinforced materials, iron, copper, brass, bronze, and other alloys. Place them in an open environment under tin shade. Observe the corrosive properties over a period of four weeks. Record your observations. Discuss the findings with your teacher and colleagues.
7. Collect the water sample from different sources of ground and surface water (at least five). Explore the new and simplest softening and water treatment methods and perform the same at your home by creating the different assemblies and manipulative techniques at home. Determine the turbidity and pH of water (using pH paper).
8. Collection of data of various cement, glass, paints, and varnishes available in the market.
9. Compare the EMF of a given cell using different fruit juice as electrolyte. Compare the hardness of different sample water by measuring the time required for forming lather.
10. Determine the flash point and fire point of a lubricant.
11. Collect petrol from different petrol pumps and compare the extent of knocking by comparing their mileage.

c. Other Activities:

1. Seminar Topics:
 - Water Softening techniques.
 - Advantages and drawbacks of different atomic structures proposed by different scientists.
 - Properties of good lubricants.
 - Application of Nernst equation.
2. Visits: Visit nearby **Water treatment** plant. Prepare report of visit.
Visit a nearby battery shop. Prepare a report of visit.
3. Self-Learning Topics: -

M) Suggested Course Evaluation Matrix: The course teacher has to decide and use appropriate assessment strategy and its weightage in theory, laboratory and Term Work for ensuring CO attainment. The response/performance of each student in each of these designed activities is to be used to calculate **CO attainment**.

COs	Course Evaluation Matrix						
	Theory Assessment (TA)**		Term Work Assessment (TWA)			Lab Assessment (LA)#	
	Progressive Theory Assessment (PTA) Class/Mid Sem Test	End Theory Assessment (ETA)	Term Work & Self Learning Assessment			Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)
			Assignments	Micro Projects	Other Activities*		
CO-1	20%	20%	15%	-	-	20%	20 %
CO-2	20%	20%	10%	25%	-	20%	20 %
CO-3	20%	20%	15%	25%	33%	15%	20 %
CO-4	15%	15%	30%	25%	33%	15%	20 %
CO-5	25%	25%	30%	25%	34%	30%	20 %
Total Marks	30	70	20	20	10	20	30
			50				

Legend:

- *: Other Activities include self- learning, seminar, visits, surveys, product development, software development etc.
 **: Mentioned under point- (N)

#: Mentioned under point-(O)

Note:

- The percentage given are approximate
- In case of Micro Projects and End Laboratory Assessment (ELA), the achieved marks will be equally divided in all those COs mapped with total experiments.
- For CO attainment calculation indirect assessment tools like course exit survey need to be used which comprises of questions related to achievement of each COs.

N) Suggested Specification Table for End Semester Theory Assessment: Specification table represents the reflection of sample representation of assessment of cognitive domain of full course.

Unit Title and Number	Total Classroom Instruction (CI) Hours	Relevant COs Number(s)	Total Marks	ETA (Marks)		
				Remember (R)	Understanding (U)	Application & above (A)
Unit-1.0 Atomic Structure and Chemical Bonding	8	CO1	14	4	4	6
Unit-2.0 Water	8	CO2	14	4	4	6
Unit-3.0 Engineering Material	8	CO3	14	4	6	4
Unit-4.0 Chemistry of fuels and Lubricants	12	CO4	10	4	2	4
Unit-5.0 Electrochemistry	12	CO5	18	4	6	8
Total	48	-	70	20	22	28

Note: Similar table can also be used to design class/mid-term/ internal question paper for progressive assessment.

O) Suggested Assessment Table for Laboratory (Practical):

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA/ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
1.	Preparation of 250 ml of N/10 Oxalic acid Solution	CO1	30	60	10
2.	Preparation of 250ml of N/10 Sodium Carbonate Solution.	CO1	40	50	10
3.	Determination of strength of Sodium Hydroxide solution by titrating against Oxalic Acid Solution.	CO1	30	60	10
4.	Determination of the total hardness of tap water by EDTA method.	CO2	30	60	10
5.	Estimation of moisture content in given coal sample gravimetrically.	CO3	30	60	10
6.	Preparation of Barium Sulphate from Barium Chloride.	CO2	30	60	10
7.	Determination of viscosity of lubricating oil using Ostwald Viscometer	CO4	30	60	10
8.	Comparison of the effect of dilution of electrolytes on the emf of a Daniel cell.	CO5	40	50	10
9.	Determination of pH of given solution by pH meter.	CO2	40	50	10
10.	Preparation of Phenol Formaldehyde Resin (Bakelite).	CO3	40	50	10
11.	Determination of dissolved Oxygen in given sample of Water.	CO2	30	60	10
12.	Determination of pH of soil using baking soda and vinegar.	CO2	30	60	10

Legend:

PRA*: Process Assessment

PDA**: Product Assessment

Note: This table can be used for both end semester as well as progressive assessment of practical. Rubrics need to be prepared by the course teacher for each experiment/practical to assess the student performance.

- P) Suggested Instructional/Implementation Strategies:** Different Instructional/ Implementation Strategies may be appropriately selected, as per the requirement of the content/outcome. Some of them are Improved Lecture, Tutorial, Case Method, Group Discussion, Industrial visits, Industrial Training, Field Trips, Portfolio Based Learning, Role Play, Live Demonstrations in Classrooms, Lab, Field Information and Communications Technology (ICT) Based Teaching Learning, Blended or flipped mode, Brainstorming, Expert Session, Video Clippings, Use of Open Educational Resources (OER), MOOCs etc.

Q) List of Major Laboratory Equipment, Tools and Software:

S. No.	Name of Equipment, Tools and Software	Broad Specifications	Relevant Experiment/Practical Number
1.	Electronic balance	Scale range of 0.001g to 500g. Pan size 100 mm; response time 3-5 sec.; power requirement 90-250 V, 10 watt.	1,2,3,5,6,7,8,9
2.	Electric oven	Inner size 18"x18"x18"; temperature range 100 to 250° C. with the capacity of 40lt.	5
3.	Ostwald Viscometer	Size 120x1 mm (length x internal diameter) Overall, Height 237 mm Material- Glass	7
4.	Digital pH Meter	Type: Microcontroller Based, Display: LED / LCD / Touch Screen, 3 digits, Calibration: up to 3 points with auto buffer, pH Range (pH): 0.00 to 14.00, +/- 0.05, Power Requirements: 230 V +/- 10, 50 Hz AC, Modes: pH mV- C, Temperature Compensation Type: Automatic, Temperature Compensation Range (Degree C): 0 to 100, Temperature Accuracy (Degree C): +/- 0.3, Resolution (pH): 0.01	9,12

R) Suggested Learning Resources:

(a) Books:

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
1.	Engineering Chemistry	Jain & Jain	Dhanpat Rai Publishing Co.(P) Ltd., New Delhi, 2015, ISBN: 93-521-6000-2
2.	A Textbook of Engineering Chemistry	Dr S. S. Dara & Dr S. S. Umare	S. Chand & Co.(P) Ltd., New Delhi, 2014, ISBN:81-219-0359-9
3.	Textbook of Chemistry for Class XI & XII (Part-I & II)	NCERT	NCERT, New Delhi, 2017-18, Class-XI, ISBN: 81-7450-494-X (part-I), 81-7450-535-O (part-II), Class-XII, ISBN: 81-7450-648-9 (part-I), 81-7450-716-7 (part-II)
4.	Engineering Chemistry	Shikha Agarwal	Cambridge Uni. Press, New Delhi, 2019, ISBN: 978-1-108-72444-9
5.	Understanding Chemistry	C.N.R. Rao	World scientific publishing Co., 2009, ISBN: 9789812836045
6.	Engineering Chemistry	Dr. Vikram, S.	Wiley India Pvt. Ltd., New Delhi, 2013, ISBN: 9788126543342
7.	Applied Chemistry Laboratory Practices, Vol. I & II	Dr. G.H. Hunger & Prof. A.N. Pathak.	NITTTR, Chandigarh, Publication, 2013-14
8.	Chemistry for Engineers	Rajesh Agnihotri	Wiley India Pvt. Ltd., 2014, ISBN: 9788126550784

9.	Fundamental of Electrochemistry	V. S. Bagotsky	Wiley International N. J.,2005, ISBN: 9780471700586
10.	Applied Chemistry with Lab manual	Anju Rawley Devdatta V. Saraf	Khanna Book Publishing Co. (P) Ltd. New Delhi, 2021, ISBN- 978-93-91505-44-8.

(b) Online Educational Resources:

1. www.chemguide.co.uk/atommenu.html (Atomic structure and chemical bonding)
2. www.visionlearning.com (Atomic structure and chemical bonding)
3. www.chem1.com (Atomic structure and chemical bonding)
4. <https://www.wastewaterelearning.com/elearning/> (Water Treatment)
5. www.capital-refractories.com (Metals, Alloys, Cement, and Refractory Materials)
6. <https://iksindia.org>
7. www.em-ea.org/guide%20books/book-2/2.1%20fuels%20and%20combustion.pdf (Fuel & Combustion)
8. www.chemcollective.org (Metals, Alloys)
9. www.wqa.org (Water Treatment)
10. [PhET: Free online physics, chemistry, biology, earth science and math simulations \(colorado.edu\)](http://phet.colorado.edu)
11. <https://www.ancient-origins.net/history-famous-people/indian-sage-acharya-kanad-001399>
12. [Courses: NPTEL](http://courses.nptel.ac.in)
13. [Virtual Labs \(vlabs.co.in\)](http://virtual-labs.vlabs.co.in)
14. olabs.edu.in
15. [Khan Academy | Free Online Courses, Lessons & Practice](https://www.khanacademy.org)

Note: Teachers are requested to check the creative commons license status/ financial implications of the suggested, online educational resources before use by the students.

(c) Others:

1. Learning Packages.
2. Lab Manuals.
3. Manufacturers' Manual
4. Users' Guide

- A) **Course Code** : **2441103 (T2441103/P2441103/ S2441103)**
- B) **Course Title** : **Fundamentals of Library and Information Science**
(CE, CSE, AIML, ME, ME (Auto), ELX, ELX (R), MIE, FTS, CRE, CHE, TE, CACDDM, GT, RE)
- C) **Pre- requisite Course(s)** :
- D) **Rationale** :

Fundamentals of Library & Information Science deals with the basics of Library as a system as well as a service. It gives broad over view of the services provided by the library as an essential part of the education system both formal and informal educations. It helps in overall personality development and quality of life of a human being. It also helps in the preservation and conservation of cultural heritage of a country. The information as resource is very crucial for the development of a country.

- E) **Course Outcomes (COs):**
After completion of the course, the students will be able to-

CO - 1 Describe basic concepts of library as a system and various services provided by the library.

CO-2 Understand librarianship as a profession and Get familiarized with various types of library.

CO-3 Recognize five laws of library science enunciated by Dr. Rangnathan, which acts as the base for all the library services.

CO-4 Enumerate names of different department of library.

CO-5 Get familiarize with various Housekeeping works of the library e.g, Acquisition, Technical processing, Book binding, Bill processing, Shelf rectification, Circulation etc.

CO-6 Understand library Co-operations- Resource Sharing, Interlibrary loan (ILL), Co-Operating Cataloguing, Centralized Procurement etc.

- F) **Suggested Course Articulation Matrix (CAM):**

Course Outcomes (COs)	Programme Outcomes(POs)							Programme Specific Outcomes* (PSOs)	
	PO-1 Basic and Discipline Specific Knowledge	PO-2 Communication Skills	PO-3 Professionalism	PO-4 Problem Solving	PO-5 Digital Working Skills	PO-6 Awareness about ethical practices	PO-7 Life Long Learning	PSO-1	PSO-2
CO-1	1	2	-	1	-	-	-		
CO-2	1	1	2	1	-	1	-		
CO-3	1	2	1	2	-	-	1		
CO-4	1	3	1	-	-	-	1		
CO-5	1	1	2	1	1	1	-		
CO-6	1	2	1	1	1	1	1		

Legend: High (3), Medium (2), Low (1) and No mapping (-)

* PSOs will be developed by respective programme coordinator at institute level. As per latest NBA guidelines, formulating PSOs is optional

G) Teaching & Learning Scheme:

Course Code	Course Title	Scheme of Study (Hours/Week)					
		Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
		L	T				
2441103	Fundamentals of Library & Information Science	03	-	04	02	09	06

Legend:

CI: Classroom Instruction (Includes different instructional/implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem based learning etc. to deliver theoretical concepts)

LI: Laboratory Instruction (Includes experiments/practical performances /problem-based experiences in laboratory, workshop, field or other locations using different instructional/Implementation strategies)

Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.

TW: Term Work (includes assignments, seminars, micro projects, industrial visits, any other student activities etc.)

SL: Self Learning, MOOCs, spoken tutorials, online educational resources etc.

C: Credits = (1 x CI hours) + (0.5 x LI hours) + (0.5 x Notional hours)

Note: TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of teacher to ensure outcome of learning.

H) Assessment Scheme:

Course Code	Course Title	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
		Theory Assessment (TA)		Term Work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
		Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2441103	Fundamentals of Library & Information Science	30	70	20	30	20	30	200

Legend:

PTA: Progressive Theory Assessment in class room (includes class test, mid-term test and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro projects, industrial visits, self-learning, any other student activities etc.

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignment, micro project, seminar and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/ presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria of internal as well as external assessment may vary as per the requirement of respective course. For valid and reliable assessment, the internal faculty should prepare checklist & rubrics for these activities.

i) Course Curriculum Detailing: This course curriculum detailing depicts learning outcomes at course level and session level and their attainment by the students through Classroom Instruction (CI), Laboratory Instruction (LI), Term Work (TW) and Self Learning (SL). Students are expected to demonstrate the attainment of Theory Session Outcomes (TSOs) and Lab Session Outcomes (LSOs) leading to attainment of Course Outcomes (COs) upon the completion of the course. While curriculum detailing, NEP 2020 related

reforms like Green skills, Sustainability, Multidisciplinary aspects, Society connect, Indian Knowledge System (IKS) and others must be integrated appropriately.

J) Theory Session Outcomes (TSOs) and Units: T2441103

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO 1a.</i> Define Library and write its purpose <i>TSO 1b.</i> List out various types of libraries. <i>TSO 1c.</i> Enumerate five laws of library science enunciated by Dr. S.R. Ranganathan	Unit-1 LIBRARY AND SOCIETY 1.1: Library and its Purpose. 1.2 : Types of Libraries 1.3: Five Laws of Library Science.	1,2&3
<i>TSO 2a.</i> Describe the working of a simple computer system. <i>TSO 2b.</i> Write the various application of computer in the library. <i>TSO 2c.</i> Define and describe library automation. <i>TSO 2d.</i> Introduce Library Management Software (LMS). <i>TSO 2e.</i> Give brief description of various LMS.	Unit -2 : IT APPLICATIONS IN LIBRARY 2.1: Computer Fundamentals. 2.2: Application of Computer in Library. 2.3 : Library Automation 2.4 Library management software(LMS)- e- Granthalaya, Koha etc...)	3&5
<i>TSO 3a.</i> List out the name of major departments of library. <i>TSO 3b.</i> Name the various library House Keeping Operations. <i>TSO 3c.</i> Identify Acquisition work and Technical processing work- Classification and Cataloguing. <i>TSO 3d.</i> Describe Inter Library Loan and other resources sharing works of library. <i>TSO 3e.</i> Describe main library networks of India.	Unit :-3 : LIBRARY-HOUSE KEEPING OPERATIONS 3.1: Departments of Library. 3.2: House Keeping works in Library- Acquisition, Technical Processing, and Book Binding. 3.3; Resource Sharing – ILL(Inter Library Loan), Cooperative cataloguing etc. 3.4; Library Networks - (DELNET, CALIBNET, INFLIBNET, etc. ...)	4,5&6
<i>TSO 4a.</i> Understand the basics of Classification <i>TSO 4b.</i> Differentiate between various Classification schemes. <i>TSO 4c.</i> Understand the basics of Cataloguing. <i>TSO 4d.</i> Write the names of various Cataloguing Codes. <i>TSO 4e.</i> Differentiate between AACR and CCC.	Unit-4 : KNOWLEDGE CLASSIFICATION AND CATALOGUING 4.1 : Basics of Classification 4.2 : Schemes of Classification – DDC,CC. 4.3 : Basics of Cataloguing. 4.4 : Cataloguing Codes- AACR, CCC.	1,5&6

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO 5a.</i> Write the names of various library services. <i>TSO 5b.</i> Describe the Circulation Services. <i>TSO 5c.</i> Differentiate between long range and short range reference services. <i>TSO 5d.</i> Identify the User Awareness services. <i>TSO 5e.</i> Differentiate between CAS and SDI services. <i>TSO 5f.</i> Describe the Translation service. <i>TSO 5g.</i> Identify other important services provided the library eg.-Reprography service, LSS, Abstracting and Indexing service.	Unit -5 :LIBRARY SERVICES 5.1 : Circulation Service 5.2 : Reference Service- Long range and Short range. 5.3 : User awareness Service-CAS, SDI. 5.4 : Translation Service. 5.5 : Other Services-Reprography service, literature Search Service, Abstracting Service, Indexing service etc.	2,3,4&5

Note: One major TSO may require more than one Theory session/Period.

K) Suggested Laboratory (Practical) Session Outcomes (LSOs) and List of Practical: P2441103

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
<i>LSO 1.1.</i> Write the process of obtaining books suggestion <i>LSO 1.2.</i> Draft a sample book order <i>LSO 1.3.</i> Prepare a reminder letter for non receipt books <i>LSO 1.4.</i> Write the process of order verification <i>LSO 1.5.</i> Mention the points to be observed while physical checking of books received. <i>LSO 1.6.</i> Write the steps of bill processing.	Unit:1	Acquisition 1.1 Suggestion management. 1.2 Order placing 1.3 Receipt of study materials 1.4 Order verification 1.5 Physical checking 1.6 Bill processing.	1,2&5
<i>LSO 2.1.</i> Write the names of columns of Accession register. <i>LSO 2.2.</i> Prepare the format of standard accession register. <i>LSO 2.3.</i> Do the simple book entry in the accession register <i>LSO 2.4.</i> Do the simple periodicals entry in the accession register <i>LSO 2.5.</i> Make the entry of Old books in the accession register (Retro-entry)	Unit:2	Accessioning 2.1 Introduction to Accessing 2.2 Accession register format 2.3 Familiarization with Accessing register. 2.4 Accessing of simple books 2.5 Accessing of periodicals	1&5
<i>LSO 3.1.</i> Write the names of Main classification schemes <i>LSO 3.2.</i> Write the names of volume of DDC. <i>LSO 3.3.</i> Write the name of 10 main classes.	Unit:3	Technical Processing: (A) Classification 3.1 Familiarization with Dewey Decimal Classification (DDC) 19 th Ed. 3.2 Familiarization with Colon Classification (CC) 6 th Ed.	1,3&5

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
<i>LSO 3.4.</i> Trace the DDC 19th edition for classifying simple books (single Author entry) <i>LSO 3.5.</i> Make the class number of simple books by using DDC. <i>LSO 3.6.</i> Write the names of main classes of CC. <i>LSO 3.7.</i> Trace the CC for classifying Simple books (single Author) <i>LSO 3.8.</i> Make the class number of simple books by using CC.		3.3 Tracing Practice of DDC 3.4 Tracing practice of Colon Classification(CC) 3.5 Classifying Single book titles by using DDC 19th Ed. 3.6 Classifying simple books by CCC.	
<i>LSO 4.1.</i> Write the dimension of standard Catalogue Card <i>LSO 4.2.</i> Prepare the format of the standard Catalogue Card <i>LSO 4.3.</i> Write about the Card Cabinet. <i>LSO 4.4.</i> Write the process of Cataloguing as per AACR-II. <i>LSO 4.5.</i> Write the rule of Cataloguing as per CCC. <i>LSO 4.6.</i> Make the Catalogue Card of simple books by using AACR-II <i>LSO 4.7.</i> Make the Catalogue Card of simple books by using CCC.	Unit:4	Technical processing: (B) Cataloguing 4.1 Familiarization with Card Catalogue 4.2 Filing of Catalogue Cards 4.3 Familiarization with AACR-II 4.4 Familiarization with CCC. 4.5 Making Catalogue of single Author by using AACR-II 4.6 Making Catalogue of single Author by using CCC.	1,3&5
<i>LSO 5.1.</i> Write the process of member registration. <i>LSO 5.2.</i> Write the main column/ fields of registration register. <i>LSO 5.3.</i> Prepare a standard registration form of users. <i>LSO 5.4.</i> Write the precaution to be taken for making issue/ return in the library. <i>LSO 5.5.</i> Do the book issue to the users. <i>LSO 5.6.</i> Make return entry of the received books. <i>LSO 5.7.</i> Do the Practice of Charging of shelves.	Unit:5	Circulation Management 5.1 Member registration 5.2 Charging/Issue of Books 5.3 Book reservation procedure 5.4 Discharging/Return of Books 5.5 Shelf Management	3,4&5
<i>LSO 6.1.</i> Write the importance of Preservation of Library Books. <i>LSO 6.2.</i> Write the measures to be taken for the safety and long life of Books. <i>LSO 6.3.</i> Write the names of Chemicals used in Preservation & Conservation. <i>LSO 6.4.</i> Explain the process of Fumigation. <i>LSO 6.5.</i> Explain the functioning of Fumigation Chamber.	Unit:6	Preservation & Conservation of Library materials 6.1 Introduction to PCLM 6.2 Physical maintenance 6.3 Placing chemicals in shelves 6.4 Fumigation 6.5 Familiarization with fumigation Chamber. 6.6 Fumigation practices	3&4

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
<i>LSO 7.1.</i> Write the importance of Stock Verification. <i>LSO 7.2.</i> Explain Annual Stock Verification. <i>LSO 7.3.</i> Write the methods of Stock Verification. <i>LSO 7.4.</i> Explain the Process of Stock Verification. <i>LSO 7.5.</i> Explain the process of Write Off.	Unit:7	Stock Verification 7.1 Introduction to stock verification 7.2 Annual stock verification 7.3 Familiarization with method of stock verification. 7.4 stock verification practice	2,3&6
<i>LSO 8.1.</i> Write about the Library Management Software (L.M.S). <i>LSO 8.2.</i> Write the Salient features of e-granthalaya.(Software). <i>LSO 8.3.</i> List out the modules of e-granthalaya. <i>LSO 8.4.</i> Explain the Steps of Book Acquisition in e-granthalaya. <i>LSO 8.5.</i> Do the Cataloguing & Accessioning of Simple Books in e-granthalaya. <i>LSO 8.6.</i> Do the circulation management using e-granthalaya. <i>LSO 8.7.</i> Describe OPAC.	Unit:8	Use of LMS: e-granthalaya 8.1 Introduction to e-granthalaya 8.2 Familiarization with e-granthalaya 8.3 Modules of e-granthalaya 8.4 Book Acquisition on e-granthalaya 8.5 Cataloguing and Accessioning 8.6 Circulation Management in e-granthalaya 8.7 Online search and OPAC	2&5

Note: in addition to above listed practical, students are suggested to practice all the examples covered by the teacher during theory sessions.

L) Suggested Term Work and Self Learning: S2441103 Some sample suggested assignments, micro project and other activities are mentioned here for reference.

a. **Assignments:** Questions/Problems/Worksheet/Exercises to be provided by the course teacher in line with the targeted COs.

b. **Micro Projects:**

1. Prepare a detailed worksheet on Acquisition system of Library with step wise description.
2. Prepare a detailed summary of housekeeping works of any Academic Library.
3. Make a project on e-granthalaya (LMS) with emphasis on its salient features, working of its modules, search facilities, etc. by giving at least five sample works of each module.
4. Prepare 10 simple Titles of Main Classes, Fundamental Categories and Common isolates according to Colon Classification.
5. Prepare 10 Titles according to DDC. Out of which five titles are to be classified. Simple titles with Tables add on device.
6. Make a project on DDC scheme of knowledge classification.
7. Make a project on Colon Classification scheme.
8. Prepare a Project on Cataloguing Code- AACR-II by giving 10 simple title Catalogues.
9. Prepare a Project on Cataloguing Code- CCC by giving 10 simple title Catalogues.
10. Make a detail project on Annual stock verification of Library.

c. Other Activities:

1. Seminar Topics:
1. Detailed Essay on Digital Library
2. Comparison of INB and BNB.
3. Essay on Resource sharing in the Library.

M) Suggested Course Evaluation Matrix: The course teacher has to decide and use appropriate assessment strategy and its weightage in theory, laboratory and Term Work for ensuring CO attainment. The response/performance of each student in each of these designed activities is to be used to calculate **CO attainment**.

COs	Course Evaluation Matrix						
	Theory Assessment (TA)**		Term Work Assessment (TWA)			Lab Assessment (LA)#	
	Progressive Theory Assessment (PTA) Class/Mid Sem Test	End Theory Assessment (ETA)	Term Work & Self Learning Assessment			Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)
			Assignments	Micro Projects	Other Activities*		
CO-1	10%	10%	15%	16%	16%	10%	16%
CO-2	15%	15%	15%	16%	16%	15%	16%
CO-3	25%	25%	20%	18%	18%	25%	18%
CO-4	15%	15%	15%	16%	16%	15%	16%
CO-5	25%	25%	25%	18%	18%	25%	18%
CO-6	10%	10%	10%	16%	16%	10%	16%
Total Marks	30	70	20	20	10	20	30
			50				

Legend:

*: Other Activities include self- learning, seminar, visits, surveys, product development, software development etc.

**: Mentioned under point- (N)

#: Mentioned under point-(O)

Note:

- The percentage given are approximate
- In case of Micro Projects and End Laboratory Assessment (ELA), the achieved marks will be equally divided in all those COs mapped with total experiments.
- For CO attainment calculation indirect assessment tools like course exit survey need to be used which comprises of questions related to achievement of each COs.

N) Suggested Specification Table for End Semester Theory Assessment: Specification table represents the reflection of sample representation of assessment of cognitive domain of full course.

Unit Title and Number	Total Classroom Instruction (CI) Hours	Relevant COs Number(s)	Total Marks	ETA (Marks)		
				Remember (R)	Understanding (U)	Application & above (A)
Unit-1 LIBRARY AND SOCIETY	7		10	5	4	3
Unit-2 IT APPLICATIONS IN LIBRARY	7		12	5	4	4
Unit-3 LIBRARY- HOUSEKEEPING OPERATIONS	12		16	6	4	8
Unit-4 KNOWLEDGE CLASSIFICATION AND CATALOGUING	12		16	4	3	4
Unit-5 LIBRARY SERVICES	12		16	5	5	6
Total	50	-	70	25	20	25

Note: Similar table can also be used to design class/mid-term/ internal question paper for progressive assessment.

O) Suggested Assessment Table for Laboratory (Practical):

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA/ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
1.	Acquisition 1.1 Suggestion management 1.2 Order placing 1.3 Receipt of study materials 1.4 Order verification 1.5 Physical checking 1.6 Bill processing.	CO-1,2&5	40	50	10
2.	Accessioning 2.1 Introduction to Accessioning 2.2 Accession register format 2.3 Familiarization with Accession register 2.4 Accessioning of simple books 2.5 Accessioning of periodicals	CO-1,3&5	40	50	10
3.	Technical Processing: (A) Classification 3.1 Familiarization with Dewey Decimal Classification (DDC) 19 th Ed. 3.2 Familiarization with Colon Classification (CC) 6 th Ed. 3.3 Tracing Practice of DDC 3.4 Tracing practice of Colon Classification(CC) 3.5 Classifying Single book titles by using DDC 19 th Ed. 3.6 Classifying simple books by CCC.	CO-2,4&5	40	50	10
4.	Technical processing: (B) Cataloguing 4.1 Familiarization with Card Catalogue 4.2 Filing of Catalogue Cards 4.3 Familiarization with AACR-II 4.4 Familiarization with CCC. 4.5 Making Catalogue of single Author by using AACR-II 4.6 Making Catalogue of single Author by using CCC.	CO-2,5&6	40	50	10
5.	Circulation Management 5.1 Member registration 5.2 Charging/Issue of Books 5.3 Book reservation procedure 5.4 Discharging/Return of Books 5.5 Shelf Management	CO-1,3&5	40	50	10
6.	Preservation & Conservation of Library materials 6.1 Introduction to PCLM 6.2 Physical maintenance 6.3 Placing chemicals in shelves	CO-1,2&4	40	50	10

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA/ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
	6.4 Fumigation 6.5 Familiarization with fumigation Chamber 6.6 Fumigation practices				
7.	Stock Verification 7.1 Introduction to stock verification 7.2 Annual stock verification 7.3 Familiarization with method of stock verification. 7.4 stock verification practice.	CO-2,3&5	40	50	10
8.	Use of LMS: e-granthalaya 8.1 Introduction to e-granthalaya 8.2 Familiarization with e-granthalaya 8.3 Modules of e-granthalaya 8.4 Book Acquisition on e-granthalaya 8.5 Cataloguing and Accessioning 8.6 Circulation Management in e-granthalaya 8.7 Online search and OPAC	CO-3,5&6	40	50	10

Legend:

PRA*: Process Assessment

PDA**: Product Assessment

Note: This table can be used for both end semester as well as progressive assessment of practical. Rubrics need to be prepared by the course teacher for each experiment/practical to assess the student performance.

P) Suggested Instructional/Implementation Strategies: Different Instructional/ Implementation Strategies may be appropriately selected, as per the requirement of the content/outcome. Some of them are Improved Lecture, Tutorial, Case Method, Group Discussion, Industrial visits, Industrial Training, Field Trips, Portfolio Based, Learning, Role Play, Live Demonstrations in Classrooms, Lab, Field Information and Communications Technology (ICT)Based Teaching Learning, Blended or flipped mode, Brainstorming, Expert Session, Video Clippings, Use of LMS, Open Source Software(OSS) Open Educational Resources (OER), MOOCs etc.

Q) List of Major Laboratory Equipment, Tools and Software:

S. No.	Name of Equipment, Tools and Software	Broad Specifications	Relevant Experiment/Practical Number
1.	Computer system	Processor Intel Core i5, 4 GB RAM, 15 GB free disk space	All
2.	Integrated Library Management Software(ILMS)	Software specification as per the size and type of the Libraries e.g-e-granthalaya, Koha, Libsys etc.	All

R) Suggested Learning Resources:**(a) Books:**

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
1.	Five laws of Library Science Ed 2, 1957.	Dr.S.R Ranganahan	Sharda Endowment Trust, Bangalore (Karnataka)
2.	Introduction to librarianship. Latest Ed. 7.	Gates (JK)	Mc Graw Hills publication, London, (UK)
3.	Public library system.	Ranganathan (SR) and Neelameghan (A)	McGraw Hill Education (India) Pvt.Ltd.2020, 2 nd Edition
4.	Theory of classification	Krishan Kumar	Vikas Publishing House, 1988
5.	Library and information management	Narayana, (GJ)	Prentice-Hall of India, (Delhi)

(b) Online Educational Resources:

1. www.ignou.ac.in
2. <https://egyankosh.ac.in>
3. <https://ndl.iitkgp.ac.in>
4. <https://nptel.ac.in>
5. <https://epgp.inflibnet.ac.in>

Note: Teachers are requested to check the creative commons license status/ financial implications of the suggested, online educational recourses before use by the students.

- A) **Course Code** : 2400104(T2400104/P2400104/S2400104)
- B) **Course Title** : Communication Skills (English) (Common for all Programmes)
- C) **Pre-requisite Course(s)** :
- D) **Rationale**

Communication forms a crucial element in the success of any organization or industry in the globalized economy. The global village gives due weightage to the English language and it enjoys a privileged status. Engineering students with English as a communicative language are open to many opportunities across the globe. This course will develop Listening, Speaking, Reading, and Writing Skills (LSRW) in the students for effective dissemination of their ideas, projects, patents, and research in the form of presentations, reports, research papers, memos, circulars, etc. Additionally, it will help students of diploma in engineering to present concepts and designs effectively along with writing CVs, Group Discussions, and Mock Interview sessions in placements and job recruitments. Though communication skills in SBTE, Bihar largely emphasizes to communicate effectively in English communication in Hindi is also focused to some extent at the diploma level. Effective Communication can be easily learned through Indian mythological scriptures like Bhagwat Geeta, Ramayana, Mahabharata, and others. (IKS)

- E) **Course Outcomes (COs):** After the completion of the course, teachers are expected to ensure the accomplishment of the following course outcomes by the learners. For this, the learners are expected to perform various activities related to three learning domains (Cognitive, Psychomotor, and Affective) in classroom/ laboratory/ workshop/ field/ industry.

After completion of the course, the students will be able to-

- CO-1** Communicate contextually in different situations.
- CO-2** Use Verbal Communication Effectively
- CO-3** Deploy Non-Verbal Communication Contextually.
- CO-4** Write various texts using vocabulary and correct grammar.
- CO-5** Draft effective business correspondence with brevity and clarity.

- F) **Suggested Course Articulation Matrix (CAM):**

Course Outcomes (COs)	Programme Outcomes(POs)							Programme Specific Outcomes* (PSOs)	
	PO-1 Basic and Discipline Specific Knowledge	PO-2 Problem Analysis	PO-3 Design/ Development of Solutions	PO-4 Engineering Tools	PO-5 Engineering Practices for Society, Sustainability and Environment	PO-6 Project Management	PO-7 Life Long Learning	PSO-1	PSO-2
CO-1	3	-	-	-	-	3	3		
CO-2	-	-	-	-	-	3	3		
CO-3	-	-	-	-	-	3	3		
CO-4	-	-	-	-	3	3	3		
CO-5	3	-	-	-	-	3	3		

Legend: High (3), Medium (2), Low (1) and No mapping (-)

* PSOs will be developed by respective programme coordinator at institute level. As per latest NBA guidelines, formulating PSOs is optional

G) Teaching & Learning Scheme:

Course Code	Course Title	Scheme of Study (Hours/Week)					
		Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+ SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
		L	T				
2400104	Communication Skills (English)	03	-	04	02	09	06

Legend:

CI: Classroom Instruction (Includes different instructional/implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem-based learning, etc. to deliver theoretical concepts)

LI: Laboratory Instruction (Includes experiments/practical performances /problem-based experiences in laboratory, workshop, field, or other locations using different instructional/Implementation strategies)

Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.

TW: Term Work (includes assignments, seminars, micro-projects, industrial visits, any other student activities, etc.)

SL: Self Learning, MOOCs, spoken tutorials, online educational resources, etc.

C: Credits = (1 x CI hours) + (0.5 x LI hours) + (0.5 x Notional hours)

Note: TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of the teacher to ensure the outcome of learning.

H) Assessment Scheme:

Course Code	Course Title	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
		Theory Assessment (TA)		Term Work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
		Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2400104	Communication Skills (English)	30	70	20	30	20	30	200

Legend:

PTA: Progressive Theory Assessment in class room (includes class test, mid-term test and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro projects, industrial visits, self-learning, any other student activities etc.

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignment, micro project, seminar and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/ presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria of internal as well as external assessment may vary as per the requirement of respective course. For valid and reliable assessment, the internal faculty should prepare checklist & rubrics for these activities.

II) Course Curriculum Detailing: This course curriculum detailing depicts learning outcomes at course level and session level and their attainment by the students through Classroom Instruction (CI), Laboratory Instruction (LI), Term Work (TW), and Self Learning (SL). Students are expected to demonstrate the attainment of Theory Session Outcomes (TSOs) and Lab Session Outcomes (LSOs) leading to the attainment of Course Outcomes (COs) upon the completion of the course. While curriculum detailing, NEP 2020-related reforms like Green skills, Sustainability, Multidisciplinary aspects, Society connect, Indian Knowledge System (IKS), and others must be integrated appropriately.

- J) **Theory Session Outcomes (TSOs) and Units: T2400104** The details of TSOs and units for communication in English is mentioned in Part – A while communication in Hindi is mentioned in Part – B in the following table.

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
Part -A (English) TSO1.a Define communication and its different forms. TSO1.b Explain the elements of communication with Case Studies from Bhagwat Geeta's conversation between Krishna and Arjun before the war. (IKS) TSO1.c Explain the linkages between different stages of communication with the help of a diagram. TSO1.d Apply the principles of effective communication and state two examples of communication from Ramayana (IKS) TSO1.e State eight for explaining different types of barriers to communication Case Studies from Mahabharata - the conversation between Kauravas and Pandavas in the war field (IKS) TSO1.f Identify the barriers to communication. TSO1.g Suggest the ways to overcome/minimize communication barriers.	Unit-1.0 Communication 1.1 Communication: Role, Relevance, Elements (Context-Sender-Message-Channel-Receiver-Feedback) 1.2 Process / Stages: Ideation- Encoding, Selecting Proper Channel, Transmission, Receiving, Decoding, Giving Feedback 1.3 7 Cs / Principles of Effective Communication: Considerate, Correct, Concrete, Concise, Clear, Complete. Courteous 1.4 Barriers to Communication: Physiological, Physical, Psychological, Mechanical, Semantic/Language, Cultural. Overcome/ minimize Barriers. 1.5 Case Studies from: - Bhagwat Geeta's conversation between Krishna and Arjun before the war (IKS) - Mahabharata the conversation between Kauravas and Pandavas in the war field (IKS)	CO1 CO2
TSO 2a. Distinguish between formal and informal communication Case Studies from Bhagwat Geeta and the different conversations of Krishna and Arjun during the war (IKS). TSO 2b. Illustrate the types of Formal Communication with examples. TSO 2c. Define verbal & non-verbal communication. TSO 2d. Explain the advantages of oral and written Communication. TSO 2e. Interpret non-verbal codes from Mahabharata (IKS) TSO 2f. Explain the role of tables, charts & graphs in communication. TSO 2g. Differentiate Intrapersonal and Interpersonal Communication with Case Studies TSO 2h. List the advantages and disadvantages of Group Communication.	Unit- 2.0 Types of Communication 2.1 Based on organizational structure: Formal (Vertical, Horizontal, Diagonal), Informal (Grapevine) 2.2 Based on the method of expression: Verbal-Oral & Written communication. Non-verbal communication and its Codes- Kinesics, Chronemics, Proxemics, Haptics, Vocalics/Paralanguage, Artifacts, Graphic and Visual Communication 2.3 Based on the number of people involved: Interpersonal, and Group Communication. 2.4 Case Studies from Bhagwat Geeta's different conversations with Krishna and Arjun during the war (IKS).	CO3
TSO 3a. Prepare a glossary of new words from the given texts. TSO 3b. Summarize the given texts in your own words. TSO 3c. Recognize the types of sentences in the given texts. TSO 3d. Find out idioms and phrases used in the given texts.	Unit-3.0 Reading Comprehension Comprehension, vocabulary enhancement and grammar exercises based on the reading of the following texts: Section-1 (Prose) 3.1 An Astrologer's Day by R K Narayan	CO4 CO5

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
TSO 3e. Write a short biography of the given writers. TSO 3f. Identify the figures of speech used in the given texts. TSO 3g. Classify the forms of poetry. TSO 3h. Elaborate the central idea / theme of the given poems in your own words.	3.2 Indian Civilization and Culture by M K Gandhi 3.3 The Secret of Work by Swami Vivekanand 3.4 My Struggle for an Education by Brooker T Washington Section-2 (Poetry) 3.5 Where the Mind is without Fear by R N Tagore 3.6 Ode on Solitude by Alexander Pope 3.7 Stopping by Woods on a Snowy Evening by Robert Frost 3.8 A Psalm of Life by H W Longfellow	
TSO 4a. Form new words adding prefix and suffix to the given root words. TSO 4b. Write synonyms and antonyms of the given words. TSO 4c. Use the given idioms and phrases in your own sentences. TSO 4d. Distinguish between acronym and abbreviation. TSO 4e. Prepare a list of technical jargons of your respective branch. TSO 4f. Identify the parts of speech of the specific words in the given sentences. TSO 4g. Fill in the blanks with suitable verb forms in the given sentences. TSO 4h. Transform the given sentences as directed. TSO 4i. Punctuate the given paragraphs.	Unit-4.0 Vocabulary and Grammar 4.1 Word Formation: Prefix, Suffix, Acronym 4.2 Synonyms, Antonyms, Homonyms, One Word Substitution, Idioms and Phrases 4.3 Technical Jargons -Related to the respective program 4.4 Parts of speech 4.5 Time and Tense 4.6 Transformation: Voice, Narration, Removal of 'Too', Question Tag 4.7 Punctuation	CO4, CO5
TSO 5a. Write the precis of the given passage with suitable title. TSO 5b. Draft letters and applications for the given purpose. TSO 5c. Compose E-mails, Notices, Memos, and Circulars. TSO 5d. Prepare reports of the projects of your respective branch. TSO 5e. Write a report on the events organized in your institute.	Unit-5.0 Professional Writing 5.1 Precis Writing 5.2 Business Letters / Applications 5.3 Drafting E-mails, Notices, Memos, Circulars 5.4 Report Writing: Project and Event/ Incident Report Writing	CO5
Part -B (हिंदी) TSO 1a सम्प्रेषण कौशल का अर्थ स्पष्ट कर सकेंगे. TSO 1b भाव एवं सम्प्रेषण में अंतर बता पाएँगे. TSO 1c सम्प्रेषण की प्रक्रिया का उल्लेख कर सकेंगे. TSO 1d श्रवण अविव्यक्ति, वाचन और लेखन की अवधारणा को स्पष्ट कर सकेंगे. TSO 1e सम्प्रेषण कौशल के निर्धारक तत्वों का विवेचन कर सकेंगे. TSO 1f प्रभावशाली सम्प्रेषण के सिद्धांतों का समावेश अपने वार्तालाप में कर सकेंगे.	Units-1.0: सम्प्रेषण सिद्धान्त एवं व्यवहार 1.1 सम्प्रेषण : परिचय , अर्थ एवं परिभाषा 1.2 सम्प्रेषण की प्रक्रिया एवं तत्व 1.3 सम्प्रेषण के प्रकार : औपचारिक एवं अनौपचारिक, शाब्दिक एवं अशाब्दिक 1.4 प्रभावशाली सम्प्रेषण के सिद्धांत एवं सम्प्रेषण व्यवधान कुरुक्षेत्र में श्रीकृष्ण- अर्जुन संवाद	CO1, CO2, CO3

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
	महाभारत युद्ध प्रारम्भ होने से पहले कुरुक्षेत्र में श्री कृष्ण ने अर्जुन के प्रश्नों के उत्तर देते हुए जीवन के सूत्र समझाए थे। ये उपदेश श्रीमद्भागवत गीता में मिलते	
TSO 2a तकनीकी कौशल एवं व्यवहार कौशल में अन्तर बता पाएँगे . TSO 2b व्यवहार कौशल का महत्व स्पष्ट कर पाएँगे . TSO 2c आत्म जागरूकता एवं आत्म विश्लेषण का विवेचन सोदाहरण कर पाएँगे . TSO 2d भावनात्मक बुद्धिमत्ता एवं करुणा, अनुकूलनशीलता एवं लचीलापन का विकास कर पाएँगे. TSO 2e दैनिक जीवन में अनुकूलनशीलता एवं लचीलापन को आत्मसात कर पाएँगे .	Unit-2.0: व्यावसायिक उत्कृष्टता हेतु व्यवहार कौशल 2.1 परिचय : तकनीकी कौशल एवं व्यवहार कौशल 2.2 व्यवहार कौशल का महत्व 2.3 जीवन कौशल : आत्म जागरूकता एवं आत्म विश्लेषण 2.4 वनात्मक बुद्धिमत्ता एवं करुणा, अनुकूलनशीलता एवं लचीलापन, व्यवहार कौशल का उपयोग श्रीराम केवट संवाद श्रीराम जब लक्ष्मण और सीता के साथ वन गमन के लिए प्रस्थान करते हैं तब सरयू नदी के पार उतारने लिए केवट से अनुरोध करते हैं।	CO1
TSO 3a पठित गद्यांश एवं पद्यांश से प्राप्त नयी शब्दावली विकसित कर पाएँगे TSO 3b दिए गये कहानियों, कविताओं एवं निबंधों का सारांश अपने शब्दों में लिख पाएँगे. TSO 3c दिए गये कहानियों, कविताओं एवं निबंधों में प्रयुक्त मुहावरों एवं अलंकारों को बता पाएँगे . TSO 3d कविताओं का भावार्थ स्पष्ट कर पाएँगे .	Unit-3.0: पाठ-बोध : शब्दावली परिवर्धन एवं व्याकरण अभ्यास 3.1 नमक का दरोगा, ईदगाह – मुंशी प्रेमचंद 3.2 बात (निबंध)- प्रताप नारायण मिश्र 3.3 वह प्रदीप जो दिख रहा है झिलमिल दूर नहीं है – रामधारी सिंह दिनकर 3.4 नर हो न निराश करो मन को – मैथिलीशरण गुप्त 3.5 कबीर के दोहे -काल्ह करे सो आज कर , जाति न पूछो साधू की , ऐसी वाणी बोलिए	CO4
TSO 4a अपनी शाखा से सम्बन्धित तकनीकी शब्दावली का चयन कर पाएँगे . TSO 4b पर्यायवाची एवं विलोम शब्दों से सम्बन्धित शब्दावली तैयार कर सकेंगे . TSO 4c दिये गये गद्यांशों में विराम चिह्नों का सही प्रयोग कर पाएँगे .	Unit-4.0: शब्दावली एवं व्याकरण 4.1 सामान्य शब्दावली 4.2 प्रशासनिक शब्दावली 4.3 शब्द भेद, अनेक शब्दों के लिए एक शब्द 4.4 विराम चिह्न 4.5 मुहावरें एवं कहावतें	CO4 CO5
TSO 5a दिए गये दिए गये गद्यांशों का संक्षेपण कर पाएँगे . TSO 5b विभिन्न प्रकार के पत्रों, आवेदनों, सूचनाओं, विज्ञप्तियों को लिख पाएँगे . TSO 5c अपनी शाखा से सम्बन्धित प्रतिवेदन लेखन कर पाएँगे . TSO 5d अपने संस्थान में हुए आयोजनों का प्रतिवेदन लिख पाएँगे.	Unit-5.0: लेखन कौशल 5.1 सार- लेखन 5.2 औपचारिक एवं व्यवसायिक पत्र लेखन 5.3 प्रारूप लेखन – सूचना, निविदा लेखन, प्रतिवेदन लेखन, बायोडाटा	CO5

Note: One major TSO may require more than one theory session/period.

K) Suggested Laboratory (Practical) Session Outcomes (LSOs) and List of Practical:P2400104 These practical's are common for both Part – A and Part -B.

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
LSO1.a Identify the emotions of the speakers.	1	Emotions of the speakers.	CO1
LSO2.a Interpret instructions of audio transcripts.	2	Instructions of audio transcripts.	CO1
LSO3.a Solve the language puzzles based on the audio transcript.	3	Language puzzles.	CO1
LSO4.a Repeat words on language lab software after listening to them.	4	Repetition of words	CO1
LSO5.a Summarize the excerpt in their own words.	5	Summarize the excerpt.	CO1
LSO6.a Answer the questions based on the listening excerpt	6	Listening excerpt	CO2
LSO7.a Differentiate the sounds of minimal pairs, syllables, words, etc.	7	Sounds of minimal pairs, syllables words etc.	CO2
LSO8.a Pronounce the words/ sentences correctly based on the phonetic transcription.	8	Phonetic transcription.	CO2
LSO9.a Read out the words and sentences based on stress and intonation marks.	9	Stress and intonation.	CO2
LSO10.a Apply the paralinguistic codes in verbal dialogues to show different emotions.	10	Paralinguistic Codes	CO2
LSO11.a Integrate the non-verbal codes in their verbal dialogues.	11	Non-verbal Codes	CO2
LSO12.a Correct the verbal and non-verbal presentations of their peer while giving feedback.	12	Feedback on Presentations	CO2
LSO13.a Differentiate the sounds of minimal pairs, syllables, words, etc.	13	Syllables and Words	CO2
LSO14.a Locate the dictated words from the excerpt.	14	Dictated words	CO3
LSO15.a Arrange the correct and logical sequence of the jumbled sentences.	15	Jumbled Sentences.	CO3
LSO16.a Read the given texts aloud with proper pauses and proper pronunciation.	16	Pronunciation.	CO3
LSO17.a Compare the point of view with their peers.	17	Point of view of Self and Peers	CO4
LSO18.a Identify the main ideas of the excerpt	18	Main ideas of the excerpt	CO4
LSO19.a Prepare a list of technical jargon and register specific to their program /industry.	19	Technical Jargons	CO5
LSO20.a Write the specifications of the machines/ equipment available in the workshops/labs.	20	Specifications of the machines/ equipment	CO5
LSO21.a Write a report on the projects of their respective branches.	21	Report on the Projects	CO5

L) Suggested Term Work and Self-Learning: S2400104 Some sample suggested assignments, micro-projects, and other activities are mentioned here for reference.

a. Assignments: Questions/Problems/Numerical/Exercises to be provided by the course teacher in line

with the targeted COs.

- d. Visit your institute's library/ web search and enlist the books, journals, and magazines related to your respective branches to prepare a bibliography consisting of names of the authors, titles of the books, publication, and place of publication.
- e. SWOT Analysis: Analyze yourself concerning your strengths and weaknesses, opportunities, and threats for your communication.
- f. Interview an eminent personality and write a report on it.
- g. Deliver a seminar for 10-12 minutes using PPT on the topic given.
- h. Prepare your timetable for a week and prioritize your activities.
- i. Visit any historical places/offices/farms/industries/development sites etc. near your city and prepare a report on it.
- j. Prepare a video of effective professional communication after listening to Bhagwat Geeta's conversation between Arjun and Krishna in the war field (IKS).

b. Micro Projects:

- i. Book review – students should read a book and then write their reviews about the book and present it in the class.
- ii. Interview any successful person in your locality in context with his life journey, inspiration social contribution, role model, and keys to success.
- iii. Prepare a register of technical jargon of the industry related to their specific branch.
- iv. Prepare a presentation on environmental issues of their locality with their solution.
- v. Listen to the dialogues of the conversation between Krishna and Arjun before the war for specific and effective Communication (IKS)

c. Other Activities:

1. Arrange a Blood Donation Camp in collaboration with a blood bank and prepare a communication plan for the same.
2. Organize a cleanliness campaign in your campus premises and nearby places and prepare hoardings, boards, collages, and posters for the same.
3. Organize a campaign on educational awareness in the nearby places and prepare an advertising campaign for the same.

d. Self- learning topics:

- Listen to different Conversations of Ramayana, (the Rama -Bharat conversation before going to Vanvaas) Mahabharata (Bheem and Arjun Conversation during War), and Bhagwat Geeta (discussions of Strategies before War) to develop effective communication Skills (IKS)
- Collect new words from daily newspapers.
- Observe negotiation skills in the nearby shops.
- Watch educational channels for improving English communication.

M) Suggested Course Evaluation Matrix: The course teacher has to decide and use the appropriate assessment strategy and its weightage in theory, laboratory, and Term Work for ensuring CO attainment. The response/performance of each student in each of these designed activities is to be used to calculate **CO attainment**. This matrix has been prepared considering both Part – A and Part -B.

COs	Course Evaluation Matrix		
	Theory Assessment (TA)**	Term Work Assessment (TWA)	Lab Assessment (LA)#

(Includes in Part -A & B)	Progressive Theory Assessment (PTA) Class/Mid Sem Test	End Theory Assessment (ETA)	Term Work & Self Learning Assessment			Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)
			Assignments	Micro Projects	Other Activities*		
CO-1	15%	20%	15%	20%	-	20%	20%
CO-2	10%	15%	10%	20%	25%	10%	20%
CO-3	20%	25%	15%	20%	25%	15%	20%
CO-4	25%	20%	30%	20%	25%	15%	20%
CO-5	30%	20%	30%	20%	25%	40%	20%
Total Marks	30	70	20	20	10	20	30
			50				

Legend:

*: Other Activities include self-learning, seminars, visits, surveys, product development, software development, etc.

**: Mentioned under point- (N)

#: Mentioned under point-(O)

Note:

- The percentage given are approximate
- In case of Micro Projects and End Laboratory Assessment (ELA), the achieved marks will be equally divided in all those COs mapped with total experiments.
- For CO attainment calculation indirect assessment tools like course exit survey need to be used which comprises of questions related to achievement of each COs.

N) Suggested Specification Table for End Semester Theory Assessment: Specification table represents the reflection of sample representation of assessment of cognitive domain of full course.

Unit Title and Number	Total Class room Instruction (CI) Hours	Relevant COs Number (s)	Total Marks	ETA (Marks)		
				Remember (R)	Understanding (U)	Application & above (A)
(Part - A)						
Unit-1.0 Communication Theory and Practice	5	CO1, CO2	10	3	3	4
Unit- 2.0 Types of Communication	5	CO3	8	2	2	4
Unit-3.0 Reading Comprehension	8	CO4, CO5	12	3	3	6
Unit-4.0 Vocabulary and Grammar	7	CO4, CO5	10	3	3	4
Unit-5.0 Professional Writing	7	CO5	10	3	4	3
(Part-B)						
Units-1.0: सम्प्रेषण सिद्धान्त एवं व्यवहार	2	CO1, CO2	3	1	1	1
Unit-2.0: व्यावसायिक उत्कृष्टता हेतु व्यवहार कौशल	2	CO3	3	1	1	1
Unit-3.0: पाठ-बोध :शब्दावली परिवर्धन, एवं व्याकरण अभ्यास	5	CO4, CO5	5	1	1	3
Unit-4.0: शब्दावली एवं व्याकरण	4	CO5	5	1	1	3
Unit-5.0: लेखन कौशल	3	CO5	4	2	1	1
Total	48	-	70	20	20	30

Note: Similar table can also be used to design class/mid-term/ internal question paper for progressive assessment.

O) Suggested Assessment Table for Laboratory (Practical):

S. No.	Laboratory Practical Titles	Relevant COs Number (s)	PLA/ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
1	Emotions of the Speakers.	CO1	30	60	10
2	Instructions of Audio Transcripts.	CO1	30	60	10
3	Language Puzzles.	CO1	30	60	10
4	Repetition of Words.	CO1	30	60	10
5	Summarize the Excerpts.	CO1	30	60	10
6	Listening Excerpts.	CO2	30	60	10
7	Sounds of minimal Pairs, Syllables and Words etc.	CO2	30	60	10
8	Phonetic Transcription.	CO2	30	60	10
9	Stress and Intonation.	CO2	30	60	10
10	Paralanguage Codes	CO2	30	60	10
11	Non-Verbal Codes	CO2	30	60	10
12	Verbal and Non-Verbal Presentations	CO2	30	60	10
13	Sounds of minimal pairs, syllables and words	CO2	30	60	10
14	Locate the Dictated Words	CO3	30	60	10
15	Jumbled Sentences.	CO3	30	60	10
16	Pronunciation.	CO3	30	60	10
17	Compare the Point of view with their Peers.	CO4	30	60	10
18	Main Ideas of the Excerpt	CO4	30	60	10
19	Technical Jargons	CO5	30	60	10
20	Specifications of the machines/ equipment	CO5	30	60	10
21	Report on the Projects	CO5	30	60	10

Legend:

PRA*: Process Assessment

PDA**: Product Assessment

Note: This table can be used for both end semester as well as progressive assessment of practical. Rubrics need to be prepared by the course teacher for each experiment/practical to assess the student performance.

- P) Suggested Instructional/Implementation Strategies:** Different Instructional/ Implementation Strategies may be appropriately selected, as per the requirement of the content/outcome. Some of them are Improved Lecture, Tutorial, Case Method, Group Discussion, Industrial visits, Industrial Training, Field Trips, Portfolio Based, Learning, Role Play, Live Demonstrations in Classrooms, Lab, Field Information and Communications Technology (ICT)Based Teaching Learning, Blended or flipped mode, Brainstorming, Expert Session, Video Clippings, Use of Open Educational Resources (OER), MOOCs etc.

Q) List of Major Laboratory Equipment, Tools and Software:

S. No.	Name of Equipment, Tools and Software	Broad Specifications	Relevant Experiment/Practical Number
1.	High end computers	Intel® Core™ i5-9400 (6-Core, 9MB Cache, up to 4.1GHz with Intel® Turbo Boost Technology) RAM: 8GB DDR 4 HDD: 3.5" 1TB 7200RPM SATA Hard Drive OS: Windows 10 Pro 64bit OEM License Other ports: Gigabyte LAN card	1 to 21
2.	Language Lab software	Teacher console supporting audio-visual language lab	1 to 21
3.	Printer	LaserJet printer	1 to 21
4.	Head Phones with microphones	Logitech H111 wired on headphones	1 to 21
5.	Computer Furniture	Computer Desk, chair	1 to 21
6.	Smart Projector	Standard Specification	1 to 21

R) Suggested Learning Resources:

(c) Books:

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
1.	Communication Skills in English (AICTE Prescribed Text Book)	Dr. Anjana Tiwari	Khanna and Khanna, New Delhi
2.	Business Communication	Dr. Nishith Rajaram Dubey, Anupam Singh	Publisher: Indra Publishing House, 2023 ISBN- 978-93-93577-69-6
3.	Communication Skills	Sanjay Kumar & Pushap Lata	Oxford University Press, India
4.	Employability Skills	Dr. Nishith Rajaram Dubey, Anupam Singh	Indra Publishing House, 2023 ISBN - 978-93-93577-68-9
5.	Technical Communication for Engineers	Shalini Verma	S. Chand
6.	English Grammar	Raymond Murphy	S. Chand
7.	British English Grammar and Composition	Dr. Ashok Kumar Singh	Student's Friends

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
8.	A Textbook of English Phonetics	T. Balasubramanian	Macmillan Publishers
9.	Thesaurus of English Words and Phrases	Roget	Simon and Schuster
10	Better English Pronunciation	J. D. O'Connor	Cambridge: Cambridge University Press, 1980
11	An English Grammar: Comprehending Principles and Rules	Lindley Murray.	London: Wilson and Sons, 1908.
12	Effective Communication Skills	Kulbhushan Kumar	Khanna Publishing House, New Delhi (Revised Edition 2018)
13	Examine your English	Margaret M. Maison	Orient Longman: New Delhi, 1964
14	Collin's English Dictionary	Harper Collins	Harper Collins, Glasgow
15	संप्रेषण कौशल	डॉ प्रवीण कुमार अग्रवाल , डॉ अवनीश कुमार मिश्रा	साहित्य भवन पब्लिकेशन : आगरा
16	आधुनिक हिंदी व्याकरण और रचना	डॉ वासुदेवनंदन प्रसाद	भारती भवन पब्लिकेशन

(b) Online Educational Resources:

1. https://www.academia.edu/37871134/COMMUNICATION_SKILLS_1ST_YR_2_pdf
2. [https://socialsci.libretexts.org/Courses/Butte_College/Exploring_Intercultural_Communication_\(Grothe\)/05%3A_Nonverbal_Processes_in_Intercultural_Communication/5.02%3A_Types_of_Nonverbal_Communication](https://socialsci.libretexts.org/Courses/Butte_College/Exploring_Intercultural_Communication_(Grothe)/05%3A_Nonverbal_Processes_in_Intercultural_Communication/5.02%3A_Types_of_Nonverbal_Communication)
3. <http://muhamadjaelani35.blogspot.com/2014/11/inquiry-letter-order-letter-complaint.html?m=1>
4. <https://www.slideshare.net/sundaredu/barriers-of-communication-53545680>
5. <https://allpoetry.com/where-the-mind-is-without-fear>
6. <https://www.poetryfoundation.org/poems/46561/ode-on-solitude>
7. <https://www.poetryfoundation.org/poems/44644/a-psalm-of-life>
8. <https://www.poetryfoundation.org/poems/42891/stopping-by-woods-on-a-snowy-evening>
9. <https://www.hindisamay.com/content/>
10. <http://kavitakosh.org/>
11. <https://bundelkhand.in/maithilisharan-gupt/nar-ho-na-nirash-karo-man-ko>
12. <https://etc.usf.edu/lit2go/92/up-from-slavery/1575/chapter-3-the-struggle-for-an-education/>
13. <https://oursmartstudy.com/english-chapter-1-class-12-pdf-download/>
14. [https://ve-iitg.vlabs.ac.in/Listening%20Skills\(Procedure\).html](https://ve-iitg.vlabs.ac.in/Listening%20Skills(Procedure).html)
15. <https://nptel.ac.in/courses/109104031>

Note: Teachers are requested to check the creative commons license status/ financial implications of the suggested, online educational resources before use by the students.

(c) Others:

1. <https://nptel.ac.in/courses/>

- A) **Course Code** : 2418104(T2418104/P2418104/S2418104)
- B) **Course Title** : Fundamental of IT System (AIML, CSE)
- C) **Pre-requisite Course(s)** :
- D) **Rationale** :

Information technology is a term that describes the entire range of information generation, storage, transmission, retrieval, and processing. Most organizations in the industry, business, non-profit organizations, and government departments now rely heavily on their information systems (IS) and information technology (IT). The information system collects, stores, and disseminates information from the organization's environment and internal operations to support organizational functions and decision-making, communication, coordination, control, analysis, and visualization. Therefore, the knowledge about the various applications areas of Information Technology including practical skills acquired through the laboratory will help students when he/she will be working with information systems.

At the end of the course, students will be able to comfortably work on computers, install and configure OS, connect it to external devices, and protect information and computers from basic abuses/attacks. This course is therefore so designed that the students will be able to apply the concepts of IT systems as and when required.

- E) **Course Outcomes (COs):** After the completion of the course, teachers are expected to ensure the accomplishment of following course outcomes by the learners. For this, the learners are expected to perform various activities related to three learning domains (Cognitive, Psychomotor and Affective) in classroom/ laboratory/ workshop/ field/ industry.

After completion of the course, the students will be able to-

Appraise computer systems and its applications for various educational, business, and industrial domain.
 Design simple digital logic circuit function using basic universal logic gates
 Configure different Operating Systems.
 Create a physical network according to the given topology and troubleshoot it.
 Classify the types of cyber-attack

F) **Suggested Course Articulation Matrix (CAM):**

Course Outcomes (COs)	Programme Outcomes (POs)							Programme Specific Outcomes* (PSOs)	
	PO-1 Basic and Discipline-Specific Knowledge	PO-2 Problem Analysis	PO-3 Design/Development of Solutions	PO-4 Engineering Tools	PO-5 Engineering Practices for Society, Sustainability, and Environment	PO-6 Project Management	PO-7 Life Long Learning	PSO-1	PSO-2
	1	-	-	-	1	2	2		
	1	1	-	2	-	-	-		
	1	2	-	2	1	-	-		
	1	2	3	-	2	2	2		
	1	1	-	-	2	2	2		

Legend: High (3), Medium (2), Low (1) and No mapping (-)

* PSOs will be developed by the respective program coordinator at the institute level. As per the latest NBA guidelines, formulating PSOs is optional

G) Teaching & Learning Scheme:

Course Code	Course Title	Scheme of Study (Hours/Week)					
		Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+ SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
		L	T				
2418104	Fundamental of IT System	03	-	04	02	09	06

Legend:

CI: Classroom Instruction (Includes different instructional/implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem based learning etc. to deliver theoretical concepts)

LI: Laboratory Instruction (Includes experiments/practical performances /problem-based experiences in laboratory, workshop, field or other locations using different instructional/Implementation strategies)

Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.

TW: Term Work (includes assignments, seminars, micro projects, industrial visits, any other student activities etc.)

SL: Self Learning, MOOCs, spoken tutorials, online educational resources etc.

C: Credits = (1 x CI hours) + (0.5 x LI hours) + (0.5 x Notional hours)

Note: TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of teacher to ensure outcome of learning.

H) Assessment Scheme:

Course Code	Course Title	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
		Theory Assessment (TA)		Term Work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
		Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2418104	Fundamental of IT System	30	70	20	30	20	30	200

Legend:

PTA: Progressive Theory Assessment in class room (includes class test, mid-term test and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro projects, industrial visits, self-learning, any other student activities etc.)

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignment, micro project, seminar and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria of internal as well as external assessment may vary as per the requirement of respective course. For valid and reliable assessment, the internal faculty should prepare checklist & rubrics for these activities.

I) Course Curriculum Detailing: This course curriculum detailing depicts learning outcomes at the course level and session level and their attainment by the students through Classroom Instruction (CI), Laboratory Instruction (LI), Term Work (SW) and Self Learning (SL). Students are expected to demonstrate the attainment of Theory Session Outcomes (TSOs) and Lab Session Outcomes (LSOs) leading to the attainment of Course Outcomes (COs) upon the completion of the course. While curriculum detailing, NEP 2020-related reforms like green skills, Sustainability, Multidisciplinary aspects, Indian Knowledge System (IKS), and others must be integrated appropriately.

J) Theory Session Outcomes (TSOs) and Units: T2418104

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO 1a.</i> Describe the anatomy of the Computer System. <i>TSO 1b.</i> List the different Input and Output devices. <i>TSO 1c.</i> Identify the different types of memory in computer systems. <i>TSO 1d.</i> Describe communication between different components of a computer. <i>TSO 1e.</i> Describe the functionalities of a computer system. <i>TSO 1f.</i> Use Internet digital Platforms	Unit-1.0 Basics of Computer System 1.1 Computer System and its Components. - Generation of Computer - Anatomy of Computer Systems - Input and output device - Motherboard - Peripherals - Backend and Front end of System Unit 1.2. Storage device in Computer System - Primary Storage - Secondary Storage 1.3. CPU Components - Register - Control Unit - ALU 1.4. Types of Bus - Address Bus - Data Bus - Control Bus 1.5 Search Engine - Introduction - Search Query - Applications of Internet Digital Platforms (BHIM, Digi-Locker, m-Parivahan, NPTEL etc.)	CO1
<i>TSO 2a.</i> Convert Binary numbers into different number systems <i>TSO 2b.</i> Classify Basic Logic gates and Universal Gates <i>TSO 2c.</i> Use basic universal logic gates to design simple digital logic circuit functions	Unit 2.0 Digital Logic and Number System 2.1 Introduction to digital computers and number system - Binary number system - Base conversions (Binary, Decimal, Hexadecimal, Octal) - Binary Coded Decimal 2.2 Basic Logic gates - AND, OR, INVERTER, XOR, XNOR - Working of Universal Gates - NAND Gate - NOR Gate	CO2
<i>TSO 3a.</i> Explain the functions and services of OS. <i>TSO 3b.</i> Explain different types of operating systems. <i>TSO 3c.</i> Write steps to Install Windows/Linux Operating System using a hypervisor. <i>TSO 3d.</i> Differentiate the licensed and freeware software.	Unit 3.0 Computer Software and Operating System 3.1 Different Types of Computer Software - Application Software - System Software - Utility Software 3.2 General features of OS - Introduction - Need, Functions, Services 3.2 Types of OS - Batch Operating System. - Multitasking/Time-Sharing OS. - Multiprocessing OS.	CO3

Major Theory Session Outcomes (TSOs)		Units	Relevant COs Number(s)
		- Real-Time OS. - Distributed OS. - Network OS. - Mobile OS 3.3 Windows & Linux Operating Systems (Installation) - Microsoft Windows OS (History Basic Features, Current State of OS) - Linux Operating System (Architecture, Components of Linux System, Kernel Mode vs User Mode, Basic Features) 3.4 Proprietary & Open-source software	
TSO.4a	Compare various computer network topologies and types of networks.	Unit.4.0 Computer Network and Internet Tools 4.1 Basic terminology of Computer Network - Network and its types (LAN, MAN, WAN) 4.2 Network Topology (Bus, Ring, Star, Mesh) 4.3 Networking Devices (Types and use) - Hub, Switch, Router, Bridge, Gateway, Modem, Repeater, Wireless Access Point, NIC 4.4 Transmission modes (Simplex, half-duplex, Full-duplex) 4.5 Modulation (Definition and Need) - Types of Analog Modulation - Types of Digital Modulation 4.6 Wired and Wireless media - Twisted -pair, - Coaxial, - Fiber Optics, - Radio - Infrared - Satellite 4.7 Internet & Intranet - URL - Internet - Intranet - Comparison between Intranet & Internet 4.8 Network Addressing (IPv4) - Internet Protocol (need, types) - Classful addressing scheme, Address space, notations, netid, hostid - Need of IPv6	CO4 and CO5
TSO.4b	Describe the functions of Networking Devices.		
TSO.4c	Classify the concepts of Modulation & Multiplexing for Digital Communication. Describe various wired and wireless media for digital communications.		
TSO.4d	Explain the use of IP addressing systems, DNS, and communication devices in the Internet and Intranet.		
TSO 5a.	Explain concepts of Information Security for Data Protection.	Unit. 5.0 Information Security 5.1 Need for Information Security - Definition of various terms of Information Security. - Cryptography - Vulnerability - Threat - Attack - Encryption - Decryption 5.2 The Principles of Security & Confidentiality, Integrity, Availability (CIA triad)	CO6
TSO 5b.	Classify various cyber-attacks.		
TSO 5c.	Describe cyber laws for data protection and IPR		

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
	5.3 Security services, Use of Firewall 5.3.1 Cyberattacks - Introduction of common types of attacks (Malware, Man-in-the-middle attack, Denial-of-service attack, SQL injection, Phishing, Password cracking.) 5.3.2 Cyber Law IT Amendment Act 2008 (Section 66 & 67)	

Note: One major TSO may require more than one Theory session/Period.

K) Suggested Laboratory (Practical) Session Outcomes (LSOs) and List of Practical: P2418104

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
<i>LSO 1.1.</i> Install device driver. <i>LSO 1.2.</i> Install given software on your system. <i>LSO 1.3.</i> Perform Registration process on digital India platform.	1.	1.1 Identify specifications of various types of computer systems available in your institute. 1.2 Install Printer, scanner driver. 1.3 Install any two freeware or open-source software/tool by using web browser 1.4 Use Digital India Platforms: BHIM, Digi-Locker, m-parivahan, NPTEL.	CO-1
<i>LSO 2.1.</i> Verify truth table of basic logic gates <i>LSO 2.2.</i> Design basic logical gates with NAND and NOR gates	2.	2.1 Using Integrated circuit (IC), verify the truth table of basic logic gates. 2.2 Verify truth table and digital logic circuits of basic logic gates with the help of NAND gate using IC. 2.3 Design digital logic circuit functions of basic logic gates with the help of the universal gate-NOR Gate using IC.	CO-2
<i>LSO 3.1</i> Identify different software in the PC <i>LSO 3.2</i> Install different operating systems on PC. <i>LSO 3.3</i> Use different Linux commands in real life.	3.	3.1 Install windows and Linux operating system 3.2 Practice of Basic UNIX Commands and various UNIX editors such as vi, ed, ex.	CO-3
<i>LSO 4.1.</i> Configure IPV4 addressing in the pc of a network <i>LSO 4.2.</i> Implement the cross-wired cable and straight-through cable using a clamping tool. <i>LSO 4.3.</i> Interpret Ping and Traceroute Output.	4	4.1 Identify the different networking devices. 4.2 Configure the IPV4 address in every computer in the computer network lab 4.3 Learn different LAN connections in the computer network lab. 4.4 Practically implement the cross-wired cable and straight-through cable using the clamping tool. 4.5 Interpreting Ping and Traceroute Output 4.6 Run Packet tracer tool	CO-4

L) **Suggested Term Work and Self-Learning: S2418104** Some sample suggested assignments, micro project and other activities are mentioned here for reference.

Assignments: Questions/Problems/Numerical/Exercises to be provided by the course teacher in line with the targeted COs.

Micro Projects:

1. Explore case study on cybercrime and prepare a brief presentation on it.
2. Conduct a market survey to identify various types of Desktop/Laptop/printer and write a report with their brief specifications.
3. Install any flavor of the Linux Operating System.
4. Prepare a report of various Network connecting devices existing at your home/ Institute Lab.
5. Identify different possible threats in the computer and present it.
6. Study cyber laws as applicable to educational institute and present it.

Other Activities:

1. Seminar Topics: - "Applications in demand"
"5G Networks ", "Mobile Networks"
2. Visit to industry for possible IT Infrastructure Installations.

Self-Learning Topics:

1. Internet-based various applications
2. Computer and Laptop specifications
3. Wireless Communications

M) Suggested Course Evaluation Matrix: The course teacher has to decide and use the appropriate assessment strategy and its weightage in theory, laboratory, and Term work for ensuring CO attainment. The response/performance of the student in each of these designed activities is to be used to calculate **CO attainment**.

COs	Course Evaluation Matrix						
	Theory Assessment (TA)**		Term Work Assessment (TWA)			Lab Assessment (LA)#	
	Progressive Theory Assessment (PTA) Class/Mid Sem Test	End Theory Assessment (ETA)	Term Work & Self Learning Assessment			Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)
			Assignments	Micro Projects	Other Activities*		
CO-1	10%	10%	10%	15%	25%	-	-
CO-2	15%	20%	20%	20%	15%	-	-
CO-3	30%	20%	10%	15%	10%	-	-
CO-4	20%	30%	30%	25%	25%	-	-
CO-5	25%	20%	30%	25%	25%	-	-
Total Marks	20	70	20	20	10	-	-
			50				

Legend:

*: Other Activities include self- learning, seminar, visits, surveys, product development, software development etc.

** : Mentioned under point- (N)

: Mentioned under point-(O)

Note:

- The percentage given are approximate
- In case of Micro Projects and End Laboratory Assessment (ELA), the achieved marks will be equally divided in all those COs mapped with total experiments.
- For CO attainment calculation indirect assessment tools like course exit survey need to be used which comprises of questions related to achievement of each COs.

N) Suggested Specification Table for End Semester Theory Assessment: Specification table represents the reflection of sample representation of assessment of cognitive domain of full course

Unit Title and Number	Total Classroom Instruction (CI) Hours	Relevant COs Number(s)	Total Marks	ETA (Marks)		
				Remember (R)	Understanding (U)	Application & above (A)
Unit-1.0 Basics of Computer System	6	CO-1	7	3	4	-
Unit 2.0 Digital Logic and Number System	10	CO-2	14	2	2	10
Unit 3.0 Computer Software and Operating System	10	CO-3	14	4	6	4
Unit.4.0 Computer Network and Internet Tools	12	CO-5	21	5	10	6
Unit. 5.0 Information Security	10	CO-6	14	6	4	4
Total	48	-	70	20	26	24

Note: Similar table can also be used to design class/mid-term/ internal question paper for progressive assessment.

O) Suggested Specification Table for Laboratory (Practical) Assessment:

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA /ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
1.	Identify specifications of various types of computer systems available in your institute	CO-1	40	50	10
2.	Install any two freeware or open-source software/tool by using Google Chrome/Mozilla Firefox/Microsoft Edge web browser.	CO-1	40	50	10
3.	Demonstrate the steps to register on the following Digital India Platforms from the following to survey Digital literacy.	CO-1	40	50	10
4.	Digital India Platforms: BHIM, Digi-Locker, m-parivahan ,NPTEL.	CO-1	30	60	10
5.	Verify the truth table of basic logic gates using Integrated circuit (IC).	CO-2	40	50	10
6.	Verify truth table and digital logic circuits of basic logic gates with the help of NAND gate using IC..	CO-2	40	50	10
7.	Design digital logic circuit functions of basic logic gates with the help of the universal gate-NOR Gate using IC	CO-2	60	30	10
8.	Categorize the different software available on your PC.	CO-3	60	30	10
9.	Install windows and Linux operating system	CO-3	60	30	10
10.	Practice of Basic UNIX Commands and various UNIX editors such as vi, ed, ex.	CO-3	50	40	10
11.	Identify the different networking devices.	CO-4	50	40	10
12.	Configure the IPv4 address in every computer in the computer network lab	CO-4	50	40	10

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA /ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
13.	Learn different LAN connections in the computer network lab.	CO-4	50	40	10
14.	Practically implement the cross-wired cable and straight-through cable using the clamping tool.	CO-4	60	30	10
15.	Interpreting Ping and Traceroute Output	CO-4	60	30	10

Legend:

PRA*: Process Assessment

PDA**: Product Assessment

Note: This table can be used for both end semester as well as progressive assessment of practical. Rubrics need to be prepared by the course teacher for each experiment/practical to assess the student performance.

- P)** Suggested Instructional/ Implementation Strategies: Different Instructional/ Implementation Strategies may be appropriately selected, as per the requirement of the content/outcome. Some of them are Improved Lectures, Tutorial, Case Methods, Group Discussions, Industrial visits, Industrial Training, Field Trips, Portfolios, Learning, Role Play, Live Demonstrations in Classrooms, Labs, Field Information, and Communications Technology (ICT)Based Teaching Learning, Blended or flipped mode, Brainstorming, Expert Session, Video Clippings, Use of Open Educational Resources (OER), MOOCs etc.

Q) List of Major Laboratory Equipment, Tools and Software:

S. No.	Name of Equipment, Tools, and Software	Broad Specifications	Relevant Experiment/Practical Number
1	Computer System (Desktop/Laptop) with minimum configuration: Operating System: Windows 11 or later version, Linux (Red Hat, Fedora, Ubuntu RAM:8 GB HDD: 512GB MS-Office :2010 (2016 preferable)	Computer System (Desktop/Laptop) with minimum configuration: Operating System: Windows 7 or later version, Linux (Red Hat, Fedora, Ubuntu RAM:2 GB (4 GB preferable), HDD: 250 GB (500 GB preferable) MS-Office :2010 (2016 preferable)	All
3	Crimping tool, RJ-45 connector (male-female), Twisted pair cable	Crimping tool, RJ-45 connector (male-female), Twisted pair cable	4
4	AND gate NOT gate, OR gate NAND gate, NOR gate, X-OR gate, Power supply, Digital IC trainer kit, connecting wires.	(IC 7408), (IC 7404), (IC 7432), (IC 7402), (IC 7400), (IC 7486)	2

R) Suggested Learning Resources:**(a) Books:**

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
1.	Computer Fundamentals Concepts Systems and Applications 8th Edition (English, Paperback,	Priti Sinha, Pradeep Sinha)	BPB Publications ISBN-13: 9788176567527 ISBN-10: 8176567523
2.	Fundamentals of Computers	E Balagurusamy	McGraw Hill Education 2009, ISBN-10 : 9780070141605 ISBN-13 : 978-0070141605
3.	Basic Principles of An Operating System	by Dr. Priyanka Rathee,	BPB Publications, 2019 ISBN-13: 9789388511711

(b) Open Educational Resources:

1. nptel.iitm.ac.in/courses/.../IIT.../lecture%2023%20and%2024.htm
2. en.wikipedia.org/wiki/Shear_and_moment_diagram
3. www.freestudy.co.uk/mech%20prin%20h2/stress.pdf
4. www.engineerstudent.co.uk/stress_and_strain.html
5. https://www.iit.edu/arc/workshops/pdfs/Moment_Inertia.pdf

Note: Teachers are requested to check the creative commons license status/ financial implications of the suggested, online educational resources before use by the students.

(c) Others:

1. Learning Packages
2. Users' Guide
3. Manufacturers' Manual
4. Lab Manuals

- A) **Course Code** : 2400006(T2400006/P2400006/S2400006)
- B) **Course Title** : Environmental Education and Sustainable Development
(Common for all Programmes)
- C) **Pre- requisite Course(s)** :
- D) **Rationale** :

Every creature depends on nature for their survival. It is therefore, not only essential but also moral responsibility of all of us to keep our environment clean & in a good condition. The global environmental issues such as clean water and sanitation, affordable & clean energy, sustainable cities & communities, etc. are best addresses through sustainable development goals. Environmental education is one of the primary activities to spread the concept of sustainability on a broader scope. In India, environmental education is considered as mandatory for all segment of education including technical education. Every creature depends on nature for their survival. It is therefore, not only essential but also moral responsibility of all of us to keep our environment clean & in a good condition. The concept of sustainable development is closely associated with environmental education to promote developments. Considering importance of environmental education and sustainable development, it became necessary to provide basics of these areas to the engineering graduates. The knowledge gained through this course will help the diploma students to take engineering decisions aligned to ensure sustainability of environment for next generations through proper protection of environment.

- E) **Course Outcomes (COs):** After the completion of the course, teachers are expected to ensure the accomplishment of following course outcomes by the learners. For this, the learners are expected to perform various activities related to three learning domains (Cognitive, Psychomotor and Affective) in classroom/ laboratory/ workshop/ field/ industry.

After completion of the course, the students will be able to-

- CO-1** Explain the importance of ecosystem for the protection of environment
- CO-2** Use relevant air & water pollution control methods to solve pollution related issues
- CO-3** Recognize relevant energy sources required for domestic & industrial application
- CO-4** Analyze the issues of climate change and its impact on sustainability
- CO-5** Apply engineering solutions/methods/legislations to reduce the activities that are harming the environment.

- F) **Suggested Course Articulation Matrix (CAM):**

Course Outcomes (COs)	Programme Outcomes(POs)							Programme Specific Outcomes* (PSOs)	
	PO-1 Basic and Discipline Specific Knowledge	PO-2 Problem Analysis	PO-3 Design/ Development of Solutions	PO-4 Engineering Tools	PO-5 Engineering Practices for Society, Sustainability and Environment	PO-6 Project Management	PO-7 Life Long Learning	PSO-1	PSO-2
CO-1	3	-	-	-	2	-	2		
CO-2	3	2	2	2	2	-	2		
CO-3	3	-	-	-	3	-	2		
CO-4	3	3	-	2	2	-	2		
CO-5	3	-	3	3	2	2	2		

Legend: High (3), Medium (2), Low (1) and No mapping (-)

* PSOs will be developed by respective programme coordinator at institute level. As per latest NBA guidelines, formulating PSOs is optional

G) Teaching & Learning Scheme:

Course Code	Course Title	Scheme of Study (Hours/Week)					
		Classroom Instruction (CI)		Lab Instruction (LI)	Notional Hours (TW+ SL)	Total Hours (CI+LI+TW+SL)	Total Credits (C)
		L	T				
2400006	Environmental Education and Sustainable Development	01	-	01	01	03	02

Legend:

CI: Classroom Instruction (Includes different instructional/implementation strategies i.e. Lecture (L), Tutorial (T), Case method, Demonstrations, Video demonstration, Problem based learning etc. to deliver theoretical concepts)

LI: Laboratory Instruction (Includes experiments/practical performances /problem-based experiences in laboratory, workshop, field or other locations using different instructional/Implementation strategies)

Notional Hours: Hours of engagement by learners, other than the contact hours for ensuring learning.

TW: Term Work (includes assignments, seminars, micro projects, industrial visits, any other student activities etc.)

SL: Self Learning, MOOCs, spoken tutorials, online educational resources etc.

C: Credits = (1 x CI hours) + (0.5 x LI hours) + (0.5 x Notional hours)

Note: TW and SL have to be planned by the teacher and performed by the learner under the continuous guidance and feedback of teacher to ensure outcome of learning.

H) Assessment Scheme:

Course Code	Course Title	Assessment Scheme (Marks)						Total Marks (TA+TWA+LA)
		Theory Assessment(TA)		Term Work & Self-Learning Assessment (TWA)		Lab Assessment (LA)		
		Progressive Theory Assessment (PTA)	End Theory Assessment (ETA)	Internal	External	Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)	
2400006	Environmental Education and Sustainable Development	15	-	10	-	10	15	50

Legend:

PTA: Progressive Theory Assessment in class room (includes class test, mid-term test and quiz using online/offline modes)

PLA: Progressive Laboratory Assessment (includes process and product assessment using rating Scales and rubrics)

TWA: Term work & Self Learning Assessment (Includes assessment related to student performance in assignments, seminars, micro projects, industrial visits, self-learning, any other student activities etc.

Note:

- ETA & ELA are to be carried out at the end of the term/ semester.
- Term Work is to be done by the students under the guidance of internal faculty but its assessment will be done **internally (40%)** as well as **externally (60%)**. Assessment related to planning and execution of Term Work activities like assignment, micro project, seminar and self-learning is to be done by internal faculty (Internal Assessment) whereas assessment of output/product/ presentation related to these activities will be carried out by external faculty/expert (External Assessment). However, criteria of internal as well as external assessment may vary as per the requirement of respective course. For valid and reliable assessment, the internal faculty should prepare checklist & rubrics for these activities.

I) **Course Curriculum Detailing:** This course curriculum detailing depicts learning outcomes at course level and session level and their attainment by the students through Classroom Instruction (CI), Laboratory Instruction (LI), Term Work (TW) and Self Learning (SL). Students are expected to demonstrate the attainment of Theory Session Outcomes (TSOs) and Lab Session Outcomes (LSOs) leading to attainment of Course Outcomes (COs) upon the completion of the course. While curriculum detailing, NEP 2020 related reforms like Green skills, Sustainability, Multidisciplinary aspects, Society connect, Indian Knowledge System (IKS) and others must be integrated appropriately.

J) **Theory Session Outcomes (TSOs) and Units: T2400006**

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
<i>TSO 1a.</i> Differentiate aquatic & terrestrial ecosystem <i>TSO 1b.</i> Explain structure of ecosystem <i>TSO 1c.</i> Compare food chain & web chain <i>TSO 1d.</i> Describe carbon, nitrogen, Sulphur & phosphorus cycle <i>TSO 1e.</i> Explain causes & effect of global warming	Unit-1.0 Ecosystem 1.1 Aquatic & Terrestrial ecosystem 1.2 Structure of ecosystem 1.3 Food chain & Food web 1.4 Carbon, Nitrogen, Sulphur & Phosphorous Cycle 1.5 Global warming – Causes & Effects	CO1
<i>TSO 2a.</i> Explain environmental pollution & its sources. <i>TSO 2b.</i> Assess the causes of water & air pollution in a given area <i>TSO 2c.</i> Explain the effects of water & air pollution on human, plant & animal <i>TSO 2d.</i> Take appropriate measures to prevent the pollution problems at city /municipal areas <i>TSO 2e.</i> Determine the pollution level in the environment at different seasons.	Unit-2.0 Air & Water Pollution 2.1 Traditional pollution issues- Air, Water, Noise 2.2 Water pollution 2.2.1 Sources of water pollution 2.2.2 Effects of water pollution 2.2.3 Control of water pollution 2.2.4 Physical & chemical standard of domestic water as per Indian Standard 2.3 Air pollution 2.3.1 Sources of air pollution 2.3.2 Air pollutants 2.3.3 Effects of air pollution on human, plant & animal 2.3.4 Air monitoring system 2.3.5 Air pollution control	CO2
<i>TSO 3a.</i> Describe various types renewable sources of energy <i>TSO 3b.</i> Explain solar energy & methods of harnessing <i>TSO 3c.</i> Explain wind energy and its impact on environment <i>TSO 3d.</i> Explain characteristics of biomass & its digestion process <i>TSO 3e.</i> Describe new energy sources & their application	Unit-3.0 Sustainability & Renewable Sources of Energy 3.1 Concept of sustainable development 3.2 Renewable sources of energy for sustainable development 3.3 Solar Energy 3.3.1 Features of solar thermal & PV system 3.3.2 Solar pond, Solar water heater, Solar dryer and Solar stills 3.4 Wind Energy 3.4.1 Current status & future prospects of wind energy	CO3

Major Theory Session Outcomes (TSOs)	Units	Relevant COs Number(s)
	3.4.2 Wind energy in India- Advantages and challenges of harnessing wind energy 3.4.3 Environmental benefits & limitations 3.5 Biomass 3.5.1 Types of Biomass energy sources 3.5.2 Energy content in Biomass of different types 3.5.3 Biogas production 3.6 Concept and advantages of hydroponics or aquaponics system to demonstrate soil less cultivation and integration of fish and plant cultivation. 3.7 Water conservation and sustainable development 3.8 New Energy Sources: Hydrogen energy, Ocean energy & Tidal energy	
<i>TSO 4a.</i> Describe impact of climate change on human life <i>TSO 4b.</i> Identify the factors contributing to climate change <i>TSO 4c.</i> Explain sustainable development goals to transform the world <i>TSO 4d.</i> Develop implementation strategies for action plan on climate change	Unit-4.0 Climate Change and Sustainable Development 4.1 Impact of Climate change 4.2 Factor contributing to climate change 4.3 Sustainable development Goals (SDGs) 4.4 Action Plan on Climate Change- India	CO4
<i>TSO 5a.</i> Identify the elements of a successful management system <i>TSO 5b.</i> Explain green building concept & its benefits <i>TSO 5c.</i> Apply 5R concept in a given building construction project <i>TSO 5d.</i> Explain various environment protection laws <i>TSO 5e.</i> Explain carbon foot-print & carbon credit	Unit-5.0 Environmental legislation and Sustainable Building Practices 5.1 Environment management system and Planning 5.2 Green Building concept 5.3 Green and sustainable building materials -5R concept 5.4 Environment protection acts, legislation and Laws 5.5 Zero carbon foot-print building for sustainable construction.	CO5

Note: One major TSO may require more than one Theory session/Period.

K) Suggested Laboratory (Practical) Session Outcomes (LSOs) and List of Practical: P2400006

Practical/Lab Session Outcomes (LSOs)	S. No.	Laboratory Experiment/Practical Titles	Relevant COs Number(s)
LSO 1.1. Use of Air pollutant analyzer to determine the air pollution level LSO 1.2. Collect air samples for pollution level detection	1.	Determination of air pollutants harming local environment	CO2
LSO 2.1 Use of Water pollutant analyzer to determine the water pollution LSO 2.2 Collect water samples for pollution level detection	2	Determine the water pollutants harming local environment	CO2
LSO 3.1 Prepare report on EIA of a given context and area. LSO 3.2 Collection of stakeholders view on effect on environment about a particular project/activity.	3.	Carry out the Environmental Impact Assessment (EIA) for a given project /activity of development	CO1 CO3
LSO 4.1 Predict of possible factors causing effects of climate change LSO 4.2 Effect of Ice melting on sea water	4.	Assessment of the impact of climate change on local environment	CO1 CO4
LSO 5.1 Elaborate the uses of sustainable building materials, the considering 3R LSO 5.2 Trace of Carbon foot print due to construction of a small building	5.	Demonstration of sustainable building materials in lab/workshop	CO2 CO5
LSO 6.1 Set up sample recycling bins in the laboratory LSO 6.2 Appreciate the importance of recycling and environmental benefits LSO 6.3 Explain the importance of 3 R	6.	Demonstration of the recycling process for the different materials such as paper, plastic etc. for waste management	CO3
LSO 7.1 Explain the process of composting LSO 7.2 disseminate the use of composting process to near and dear for soil health and fertility for generating organic food	7	Setting up composting bins in the laboratory to demonstrate the process of composting organic waste	CO3
LSO 8.1 Calculate own water footprint for daily activities LSO 8.2 Explain the importance of reducing water consumption and conserve water resources.	8	Calculation of personal water footprint for daily water usage for activities like bathing, cooking and laundry.	CO3
LSO 9.1 Explore the alternative / renewable sources of energy in day to day life	9.	Develop bio mass energy in the laboratory	CO3 CO4
LSO 10.1 Explore the alternative / renewable sources of energy in day to day life	10.	Develop solar model in the laboratory	CO3
LSO 11.1 Explore the alternative / renewable sources of energy in day to day life	11.	Develop wind turbine model in the laboratory	CO4

- L) Suggested Term Work and Self Learning: S2400006** Some sample suggested assignments, micro project and other activities are mentioned here for reference.
- a. Assignments:** Questions/Problems- Real life problem /Numerical/Exercises to be provided by the course teacher in line with the targeted COs.
1. Conduct a waste audit in your polytechnic. Categorize waste into different types such as plastic, paper, organic. Quantify the amount of each waste.
- b. Micro Projects:**
- Conduct of EIA of a project/activity such as construction of roads in the local area. Prepare a report on:
 - (a) Environmental issues in your city
 - (b) SDGs and environment related acts/laws applicable in your state and in India.
 - (c) Current-status & future-prospects of Wind Energy
 - (d) New energy sources
 - Prepare a model of rain water harvesting system to demonstrate how rainwater can be collected and stored for various purposes such as irrigation and toilet flushing.
 - Students may be asked in group to set up a small solar panel to compare the energy output under different lighting condition and angles to understand the concept of solar energy and its potential applications.
- c. Other Activities:**
1. Seminar Topics:
 - Climate change issue and problems
 - Sustainable development- Global practices
 - Factor affecting sustainability in India
 2. Visits:

Visit Pollution control Board of your city. Prepare report of visit with special comments of initiatives taken for protecting environment and ensuring sustainable development of the city.

Organize a field trip to a nearby park for the students. Students can be observed different species of the plants, animals and insects. They may be asked to prepare report on importance of biodiversity conservation.
 3. Self-Learning Topics:
 - Sustainable Development Goals
 - Climate change.
 - Pollution issues
 - Laws and legislation of environmental protection

M) Suggested Course Evaluation Matrix: The course teacher has to decide and use appropriate assessment strategy and its weightage in theory, laboratory and Term Work for ensuring CO attainment. The response/performance of each student in each of these designed activities is to be used to calculate **CO attainment**.

Cos	Course Evaluation Matrix						
	Theory Assessment (TA)**		Term Work Assessment (TWA)			Lab Assessment (LA)#	
	Progressive Theory Assessment (PTA) Class/Mid Sem Test	End Theory Assessment (ETA)	Term Work & Self Learning Assessment			Progressive Lab Assessment (PLA)	End Laboratory Assessment (ELA)
			Assignments	Micro Projects	Other Activities*		
CO-1	-	-	15%	-	-	20%	20%
CO-2	-	-	10%	25%	-	10%	20%
CO-3	-	-	15%	25%	50%	15%	20%
CO-4	-	-	30%	50%	50%	15%	20%
CO-5	-	-	30%	-	-	40%	20%
Total Marks	-	-	10	10	05	10	15
			25				

Legend:

*: Other Activities include self- learning, seminar, visits, surveys, product development, software development etc.

** : Mentioned under point- (N)

#: Mentioned under point-(O)

Note:

- The percentage given are approximate
- In case of Micro Projects and End Laboratory Assessment (ELA), the achieved marks will be equally divided in all those COs mapped with total experiments.
- For CO attainment calculation indirect assessment tools like course exit survey need to be used which comprises of questions related to achievement of each COs.

N) Suggested Specification Table for End Semester Theory Assessment: (Not Applicable)

O) Suggested Assessment Table for Laboratory (Practical):

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA/ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
1.	Determine the Air and water pollutants harming local environment	CO1	30	60	10
2.	Determine the water pollutants harming local environment	CO1	40	50	10
3.	Carry out the Assessment of Environmental Impact (EIA) for a given project /activity of development	CO1 CO3	30	60	10
4.	Assess the impact of climate change on local environment	CO1 CO4	30	60	10
5.	Demonstrate sustainable building materials in lab/workshop	CO2 CO5	30	60	10
6.	Demonstrate the recycling process for the different materials such as paper, plastic etc. for waste management	CO3	50	40	10
7.	Setting up composting bins in the laboratory to demonstrate the process of composting organic waste	CO3	50	40	10
8.	Calculation of personal water footprint for daily water usage for activities like bathing, cooking and laundry.	CO3	50	40	10
9.	Develop bio mass energy in the laboratory	CO3 CO4	30	60	10

S. No.	Laboratory Practical Titles	Relevant COs Number(s)	PLA/ELA		
			Performance		Viva-Voce (%)
			PRA* (%)	PDA** (%)	
10.	Develop solar model in the laboratory	CO3	30	60	10
11.	Develop Wind turbine model in the laboratory	CO4	40	50	10

Legend:

PRA*: Process Assessment

PDA**: Product Assessment

Note: This table can be used for both end semester as well as progressive assessment of practical. Rubrics need to be prepared by the course teacher for each experiment/practical to assess the student performance.

P) Suggested Instructional/Implementation Strategies: Different Instructional/ Implementation Strategies may be appropriately selected, as per the requirement of the content/outcome. Some of them are Improved Lecture, Tutorial, Case Method, Group Discussion, Industrial visits, Industrial Training, Field Trips, Portfolio Based, Learning, Role Play, Live Demonstrations in Classrooms, Lab, Field Information and Communications Technology (ICT)Based Teaching Learning, Blended or flipped mode, Brainstorming, Expert Session, Video Clippings, Use of Open Educational Resources (OER), MOOCs etc.

Q) List of Major Laboratory Equipment, Tools and Software:

S. No.	Name of Equipment, Tools and Software	Broad Specifications	Relevant Experiment/Practical Number
1.	Air analyzer	Air Quality Meter Product Type: Measuring Instrument Analysis Time: 2 sec to 8-hour 59 min. 59 sec Automation Grade: Automatic	1
2.	Water Analyzer	Multi-Parameter Water Testing Meter Digital LCD Multi-Function Water Quality Monitor PH/EC/TDS/Salt/S. G/CF/ORP	2
3.	Sustainable Building Materials	As per availability in the market	2,5
4.	Solar energy Panel – KT	Solar Panel Kit 5 LEDs, 2 ON/Off Switch, Wire, 2 Crocodile Clip	7
5.	Bio mass/energy installation -kit	The Bio-energy Science Kit is a great way to find out how a direct ethanol fuel cell works.	6
6.	Wind power energy -Kit	4M wind turbine kit, to demonstrate power of wind and convert it into electricity by building your own turbine.	8
7.	Ice melting demo kit	Simple bowls of different sizes	--

R) Suggested Learning Resources:

(a) Books:

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
1.	Ecology and Control of the Natural Environment	Izrael, Y.A.	Kluwer Academic Publisher eBook ISBN: 978-94-011-3390-6
2.	Renewable Energy Sources and Emerging Technologies	Kothari, D.P. Singal, K.C., Ranjan, Rakesh	PHI Learning, New Delhi, 2009 ISBN-13 - 978-8120344709
3.	Green Technologies and Environmental Sustainability	Singh, Ritu, Kumar, Sanjeev	Springer International Publishing, 2017 eBook ISBN 978-3-319-50654-8

S. No.	Titles	Author(s)	Publisher and Edition with ISBN
4.	Coping with Natural Hazards: Indian Context	K. S. Valadia	Orient Longman ISBN-10: 8125027351 ISBN-13: 978-8125027355
5.	Introduction to Engineering and Environment	Edward S. Rubin	Mc Graw Hill Publications ISBN-10: 0071181857 ISBN-13: 978-0071181853
6.	Environmental Science	Subrat Roy	Khanna Book Publishing Co. (P) Ltd. ISBN-978: 93-91505-65-3

(b) Online Educational Resources:

1. http://www1.eere.energy.gov/wind/wind_animation.html
2. http://www.nrel.gov/learning/re_solar.html
3. http://www.nrel.gov/learning/re_biomass.html
4. <http://www.mnre.gov.in/schemes/grid-connected/biomass-powercogen/>
5. <http://www.epa.gov/climatestudents/>
6. <http://www.climatecentral.org>
7. <http://www.envis.nic.in/>
8. <https://www.overshootday.org/>
9. <http://www.footprintcalculator.org/>
10. <https://www.carbonfootprint.com/calculator.aspx>

Note: Teachers are requested to check the creative commons license status/ financial implications of the suggested, online educational resources before use by the students.

(c) Others:

1. www.nptel.iitm.ac.in
2. www.khanacademy
